

# Antibody-Drug Conjugates: Design, Formulation and Physicochemical Stability

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**ABSTRACT** The convergence of advanced understanding of biology with chemistry has led to a resurgence in the development of antibody-drug conjugates (ADCs), especially with two recent product approvals. Design and development of ADCs requires the synergistic combination of the monoclonal antibody, the linker and the payload. Advances in antibody science has enabled identification and generation of high affinity, highly selective, humanized or human antibodies for a given target. Novel linker technologies have been synthesized and highly potent cytotoxic drug payloads have been created. As the first generation of ADCs utilizing lysine and cysteine chemistries moves through the clinic and into commercialization, second generation ADCs involving site specific conjugation technologies are being evaluated and tested. The latter aim to be better characterized and controlled, with wider therapeutic indices as well as improved pharmacokinetic-pharmacodynamic (PK-PD) profiles. ADCs offer some interesting physicochemical properties, due to conjugation itself, and to the (often) hydrophobic payloads that must be considered during their CMC development. New analytical methodologies are required for the ADCs, supplementing those used for the antibody itself. Regulatory filings will be a combination of small molecule and biologics. The regulators have put forth some broad principles but this landscape is still evolving.

**KEY WORDS** antibody-drug conjugate · formulation · linker · payload · physicochemical characteristics

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## ABBREVIATIONS

AcBut	4-(4'-acetylphenoxy) butanoic acid
AcPac	3-acetylphenyl acetic acid
ADA	Anti-drug antibody
ADC	Antibody-drug conjugate
ADCC	Antibody-dependent cell-mediated cytotoxicity
AF	Auristatin F
AUC	Area under the curve
bac	Bromoacetamido moiety
BLA	Biologics License Application
C2-keto-Gal	2-acetonyl-2-deoxy-galactose
Calich	N-acetyl gamma calicheamicin
CD	Circular dichroism spectroscopy
CDC	Complement-dependent cytotoxicity
CHO	Chinese hamster ovary cells
DAR	Drug antibody ratio
DM	Disulfide-containing maytansinoids
DMA	Dimethyl acid linker
DMH	Dimethyl hydrazide linker precursor
Doe	Dolaphenine
DP	Drug product
DS	Drug substance
DSC	Differential scanning calorimetry
DSI	Drug substance intermediate
DTNB	Dithionitrobenzoate
DTT	Dithiothreitol
ELISA	Enzyme-linked immunosorbent assay
ESI-MS	Electrospray ionization mass spectroscopy
EVA	Ethylvinyl acetate
Fab	Fragment antigen-binding
FcRn	Neonatal Fc receptor for IgG
FcγR	Fc gamma receptor for IgG
FTIR	Fourier transform infrared spectroscopy

GalNAz	<i>N</i> -acetyl-azido-galactosamine	TCEP	Tris(2-carboxyethyl)phosphine
GlcNAc	<i>N</i> -acetyl-D-glucosamine	TDC	THIOMAB drug conjugate
HAHA	Human antibody-human antibody immune response	T-DM1	Ado-trastuzumab emtansine (Kadcyla)
HC	Heavy chain of IgG	UF/DF	Ultrafiltration/Diafiltration
HDPE	High density polyethylene	vc	Valine-citrulline cleavable linker
HIC	Hydrophobic interaction chromatography		
HIPS	Hydrazine-iso-Pictet-Spengler ligation		
huFcγR	Human Fc gamma receptor for IgG		
IND	Investigational New Drug		
IV	Intravenous		
LC	Light chain of IgG		
LDPE	Low density polyethylene		
mAb	Monoclonal antibody		
MALDI-TOF	Matrix Assisted Laser Desorption/Ionization Time of Flight		
mc	Maleimidocaproyl linker		
MCC	[ <i>N</i> -maleimidomethyl]cyclohexane-1-carboxylate		
MMAD	Monomethyl auristatin D		
MMAE	Monomethyl auristatin E		
MMAF	Monomethyl auristatin F		
mTG	Microbial enzyme transglutaminase		
MTX	Methotrexate		
nAF	Oxamine-auristatin F containing a short polyethylene glycol spacer		
NEM	<i>N</i> -ethylmaleimide		
nnAA	Non-native amino acids		
OBP	Office of Biotechnology Products		
ONDQA	Office of New Drug Quality Assessment		
pabc	<i>p</i> -aminobenzyloxycarbonyl		
pAcPhe	<i>p</i> -acetylphenylalanine		
PBD	Pyrrolobenzodiazepine		
PD	Pharmacodynamics		
PEG	Polyethylene glycol		
PETG	Polyethylene terephthalate glycol		
PK-PD	Pharmacokinetic-pharmacodynamic		
PTFE	Polytetrafluoroethylene		
RP-HPLC	Reverse phase high performance liquid chromatography		
scFv	Single-chain variable fragment		
SEC-HPLC	Size exclusion column high performance liquid chromatography		
SMCC	Succinimidyl-4-( <i>N</i> -maleimidomethyl)cyclohexane-1-carboxylate		
SPDP	<i>N</i> -succinimidyl-3-(2-pyridyldithio)propionate		

## INTRODUCTION

Antibody therapeutics have had considerable success for patients with solid tumors and hematological malignancies. In the last two decades, the US Food and Drug Administration has approved about a dozen antibody-based therapies to treat certain cancer types (1) and there are currently a large number of ongoing early and late stage clinical trials for additional therapies. Trastuzumab (Herceptin™) is one such example from Genentech/Roche Ltd. for solid tumors. Trastuzumab works by attaching to and blocking specific antigens (HER2/neu protein) which when activated promotes cell growth. Some breast cancer cells overexpress the HER2 gene and cause the cancer to grow and spread. When Trastuzumab attaches to the antigen it blocks the chemical signals that can stimulate uncontrolled growth. Even though there has been great success with antibody therapeutics due to their binding specificity to cells expressing their cognate antigen, antibodies alone can have some disadvantages which include, but are not limited to, low cytotoxicity and weak penetration into tumors (2). On the other hand, systemically administered anti-cancer cytotoxics, although highly potent, reach tumor cells randomly as they circulate. They also diffuse readily into both tumor cells and normal tissue. To achieve sufficient therapeutic effect at target, they may have to be dosed up to overcome lack of specificity to target. They are therefore associated with significant toxicity and non-target effects, limiting their utility and efficacy. One approach used to address these limitations of antibodies and small molecule cytotoxics is to combine them to create antibody-drug conjugates, utilizing the targeting ability of the antibody with the high cell-killing ability of the cytotoxic. Conjugation of the cytotoxic to the antibody reduces its systemic toxicity while still conjugated and in circulation. Antibody-drug conjugates are therefore highly potent targeted-therapy molecules, designed to specifically bind to cancer cells, minimize side effects and provide a wider therapeutic window (3,4). The construct created by combining the specificity of each antibody for its target antigen and the cytotoxicity of a small molecule drug, extends the therapeutic potential of both the antibody and the cytotoxic. With the first approval of an antibody-drug conjugate (ADC) being gemtuzumab ozogamicin (Mylotarg®) in 2001 and the more recent approvals of brentuximab vedotin in 2011 (Adcetris™) (5) and ado-trastuzumab emtansine in 2013 (Kadcyla™, T-DM1) (6) having validated the technology, drug companies

are actively pursuing ADCs with approximately 30 in clinical trials for solid tumors and hematological indications. Although gemtuzumab ozogamicin was subsequently withdrawn from the US market in 2010 (while currently undergoing further clinical trials), it is still approved and marketed in Japan.

ADCs are designed to selectively dispatch highly potent cytotoxic anticancer drugs to cancer cells while, ideally, leaving the healthy tissue relatively unaffected. Tumor-targeting monoclonal antibodies can be conjugated to cytotoxic anticancer drugs or payloads using a variety of linker chemistries with labile bonds. On injection, the antibody binds to specific antigens or receptors at the surface of the cancer cell. The cancer cell internalizes the entire ADC, the linker is then degraded and the cytotoxic payload is released. The exact mechanism of release will depend upon the type of linker used. These cytotoxic payloads can then kill the tumor cells through their established cytotoxic mechanisms. This targeted delivery maximizes its antitumor effect while minimizing damage to healthy tissue as would have occurred in traditional delivery of cytotoxic drugs.

The development of a successful ADC requires knowledge and understanding of the underlying biology as well as of the chemistry in play. The analytical complexity is significantly magnified relative to that for a mAb. Addressing manufacturing and regulatory issues involves consideration of biotherapeutics as well as small molecules perspectives. This review covers some of the design and development aspects of ADCs with a focus on biophysical and chemical stability of the ADC drug product. We cover the basics of the design of ADCs by examining the requirements for each element of the ADC construct. The ADC construct has significant impact on its physicochemical stability and therefore the formulation and analytical challenges encountered. The review then concludes with a consideration of quality and some key regulatory aspects as they relate to ADCs. To our knowledge, these aspects have not been reviewed together before, and this review therefore provides a good introduction to a scientist embarking on developing ADCs for the first time. Some related reviews published recently include (7–9).

Targeting of anticancer drugs using antibodies is also being attempted in modalities such as immunotoxins and immunoliposomes but is outside the scope of this review. Some excellent references are available (2,10,11). Other technologies for improved delivery of anticancer drugs based on nanotechnology, colloidal systems, polymers *etc.* are covered in directed reviews (see *e.g.*, (12–15)).

## DESIGN OF ADCS

The building blocks of an ADC comprise the monoclonal antibody, the linker and the payload (Fig. 1). Early attempts

to create ADCs using murine or chimeric antibodies conjugated to chemotherapeutic agents did not achieve enough localization in the target tumor to be efficacious. The antibodies also resulted in immunogenicity, the antigens targeted were not selective enough resulting in off-target toxicity, and the drugs were not potent enough to be able to kill the tumors [summarized in (16)]. Some of these shortcomings were addressed in the development of gemtuzumab ozogamicin which comprises a humanized antibody and a highly potent cytotoxic derivative of calicheamicin. Early clinical results were promising (17) but some later trials showed no additional benefit of the ADC in combination with chemotherapy (18). Gemtuzumab ozogamicin was withdrawn from the U.S. market in 2010 (although it is still marketed in Japan and undergoing further clinical trials). The knowledge gained from these early experiences, as well as the greatly expanded understanding of the chemistry and biology underlying the design and function of ADCs has led to a surge in the number of programs under development.

The first design element of an ADC is the mAb. Selection of the mAb is based on the following requirements, which must be retained to a very large extent even after the conjugation process. Most if not all of these properties are superior in humanized or human mAbs in contrast to murine or chimeric mAbs.

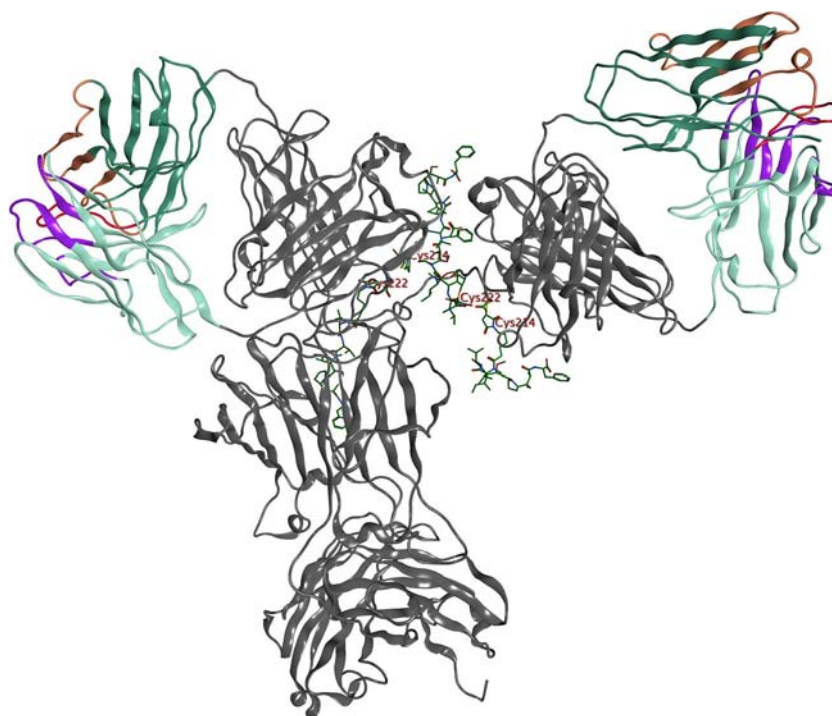
Elements of ADC's:

- High specificity and binding affinity to target antigen on tumor cells
- Efficient receptor-mediated internalization and intracellular trafficking
- Low immunogenicity potential
- Long circulation time
- Potential desirability for effector function mediated antitumor activity towards the target cells such as antibody-dependent cell-mediated cytotoxicity (ADCC) or complement-dependent cytotoxicity (CDC)
- Retention of above properties after conjugation

The second major design element of an ADC is the cytotoxic payload. The requirements for the payload are:

- High potency to be effective at concentrations achievable in the cell by the mAb-based delivery mechanism
- Selectivity in killing (proliferating) target cells
- Low immunogenic potential
- Compatibility with conjugation process (linker attachment site, process chemistry)
- Chemical stability in circulation
- Active in linked form (non-cleavable linker systems) or in free form (cleavable linker systems) based on design of ADC

**Fig. 1** A structural model of an antibody-drug conjugate showing 4 auristatins (mc-MMAF) attached to the upper hinge region (LC:C214 and HC:C222) on a human IgG1 mAb. The stick representations are provided for the linker and payload while the mAb is indicated by the ribbon diagram.



The third major design element of an ADC is the “linker”. The requirements for the linker are:

- Presence of suitable site for attachment to antibody as well as payload
- Presence of complementary attachment site on antibody to achieve desired drug loading per antibody molecule
- Stability to process chemistry and product storage
- Stability in circulation
- Compatibility with desired mechanism of action (cleavable after internalization)

Each of these elements is reviewed in greater detail below.

## Monoclonal Antibody

### Antibody Type

Based on early experiences with immunogenicity from murine mAbs as well as the current state of the art in mAb technology, the majority of ADCs in development comprise humanized or human mAbs with a few chimerics. Also, IgG<sub>1</sub>s are the most common but a few IgG<sub>2</sub> and IgG<sub>4</sub> (hinge stabilized) are also in development (19,20).

Recently approved brentuximab vedotin is based on a chimeric IgG<sub>1</sub> (cAC10) against CD30 receptors on tumor cells in Hodgkin's lymphoma and Anaplastic Large Cell Lymphoma. The package insert reports an incidence of persistent anti-drug antibody (ADA) (7%) and transient ADA (30%) in its

Phase 2 trials. The ADA is directed against the antibody component in both persistent and transient cases. Of the persistently positive patients, 1% developed adverse reactions consistent with infusion reactions and discontinued treatment. Overall, patients with persistent positive ADA experienced a higher level of infusion related reactions. Sixty-two percent of ADA patients had neutralizing ADA. However, the impact of ADA on safety and efficacy is not known (21).

T-DM1 is based on the previously approved humanized anti-HER2 IgG<sub>1</sub> antibody trastuzumab and reports 5.3% of patients developing ADA at one or more post-dose timepoints (although true incidence may be masked by presence of the drug itself in the serum). Clinical significance of the ADA was not reported (22). In contrast, the human antibody-human antibody (HAHA) response was detected in only one patient out of 903 for trastuzumab (23). Although incidence of ADA cannot be compared across products due to differences in assays, considering the close relationship between these latter two products, it may be speculated that the (significantly) higher ADA with T-DM1 may be related to structural perturbations caused by the conjugation.

ADCs for oncology indications are not intended for chronic use and so the role of immunogenicity may be important only if function is impacted. Considering the state of the art in monoclonal antibody therapy, it is unlikely that a murine antibody would be brought forward. Therefore, the question of antibody selection boils down to sequence and structural features for reasons elaborated below (24).

## Affinity

Monoclonal antibodies used in cancer therapy function through a variety of mechanisms; receptor down regulation, induction of apoptosis through inhibition of receptor-linked signaling pathways either by targeting the receptor or ligand, or effector functions (ADCC, CDC) (25,26). Internalization ability is generally not a necessary requirement in these modes of action. A majority of the current ADC approaches are, however, based on tumor cell-surface (antigen) targeting followed by internalization to deliver the payload inside the cell. An ADC design parameter that can impact the function is binding affinity. In general, the highest affinity is considered the most desirable but evidence suggests that a fast on-rate and a slow off-rate may not be the optimum (25). High binding affinity may lead to fast binding to the peri-vasculature with subsequent internalization and catabolization, and thus limit diffusion deeper into the tumor. Monoclonal antibodies, with long circulation half-lives as well as neonatal Fc receptor (FcRn)-mediated lysosomal retrieval, can achieve significant accumulation in the tumor cell without resorting to very high binding affinities (25,27). Goldmacher and Kovtun (19) also note that an intravenous (IV) administered ADC generates high initial blood concentrations (of the order of 0.1  $\mu\text{M}$  or more) while typical  $K_d$  of the antibody ranges in the nM or lower. Thus, if the concentration of ADC at the tumor site is same as that in the blood, the antigen would be saturated, irrespective of the  $K_d$ . Schmidt and Wittrup's mathematical model suggests that IgGs of  $K_d$  as low as  $10^{-7}$  M can achieve substantial accumulation in tumors (27).

## Isotype and Glycosylation

It is generally accepted that only careful *in vitro/in vivo* testing can determine which isotype fits a specific therapeutic purpose (28). However, for ADCs in oncology, antibodies of the IgG<sub>1</sub> isotype are the most common in the development pipelines due to expected additional benefit (in anti-tumor activity) of the immune effector functions ADCC and CDC, which are typically weak to negligible in IgG<sub>2s</sub> or IgG<sub>4s</sub>. [For reference, Kadcyla displays ADCC activity similar to trastuzumab, while Adcetris has minimal ADCC and no detectable CDC, although both are IgG<sub>1s</sub> (29,30)]. A contrary argument however suggests that binding to effector cells could interfere with targeting and internalization of the ADC by tumor cells, thus reducing its overall efficacy. It is unlikely that multiple isotypes would be taken to the clinic to compare efficacy and enable selection of the optimum isotype. There are few published studies on comparison of isotypes in the function of ADCs. McDonagh *et al.* (31) prepared a variety of anti-CD70 ADCs with IgG<sub>1</sub>, IgG<sub>2</sub> and IgG<sub>4</sub> antibodies and variants. The IgG<sub>1s</sub> and IgG<sub>4s</sub> were engineered to remove Fc gamma receptor (FcγR) binding while the IgG<sub>4</sub> was also mutated to impair

complement protein binding and to stabilize the hinge. Optimum conjugation conditions were found to be different for the different isotypes, ascribed to the differences in hinge-region sequence as well as disulfide bonding patterns. Conjugation to cysteines was shown to not impair FcγR binding (when not engineered away). All conjugates were found to have cytotoxicity activity *in vitro* in cell lines expressing the CD70 antigen, but potency varied depending on the cell line as well as isotype/engineered variant. Differences in pharmacokinetics of the ADCs were observed in mice as well as in maximum tolerated dose. Constructs which had improved AUC had somewhat better *in vivo* antitumor activity, although differences depended on the model. FcγR binding was not found to be correlated to improvement in therapeutic index. Thus, no general rule can be extracted from this study to address the question of preference of isotype, all the more so because of differences in human and mouse immune systems.

Although glycosylation plays an important role in the function (esp. effector function) and stabilization of antibodies (32), not much has been reported beyond this on any other roles in ADCs. Nevertheless, significant understanding of the molecular interactions between IgG-Fc and effector ligands has been generated. For maximal ADCC activity, a non-fucosylated IgG<sub>1</sub> format has been proposed to be optimal (33). An example of glycoengineering for enhanced ADCC is the recently approved antibody obinutuzumab (Gazyva™) which displays bisected non-fucosylated sugars (34). The PK of most commonly occurring mAb glycoforms is not strongly dependent on the glycoforms (35), which suggests that the ADCs would also be insensitive to the same. The two main receptors involved are the asialoglycoprotein and the mannose receptors. The asialoglycoprotein receptor recognizes complex N-linked glycan structures with missing sialic acid and terminating in galactose (G1F, G2F, G1F-GlcNAc), whereas the mannose receptor recognizes high mannose and terminal *N*-acetylglucosamine (G0F) structures. Consistency of glycoforms is therefore important, especially since binding to the mannose receptor leads to faster clearance (36–38).

## Scaffold

Along with isotype, there is the choice of the antibody scaffold to use in the ADC. Full-length antibodies remain the primary scaffold in the current development pipeline. Full-length antibodies have long circulation times which can be an advantage from a targeting perspective, allowing greater chance for reaching the tumor. However, a disadvantage of long residence time in circulation can be the enhanced risk of degradation of linker/payload, leading to non-specific toxicity. This leaves open the possibility of using antibody fragments, such as single-chain variable fragments (scFvs), fragment antigen-binding (Fab) or even other proteins to serve as carriers. Schmidt and Wittrup's model (27) suggests that small proteins



will have high penetration rates into tumors, which when combined with high binding affinity and slow off-rates, can result in substantial accumulation in tumors. But the model also suggests an uptake minimum around ~25 kDa which implies a lack of benefit from scFvs (~27 kDa), which furthermore have monovalent binding and thus lower avidity than a divalent mAb (or Fab<sub>2</sub>). Small scFV fragments (~27 kDa) will have efficient penetration and distribution but also poor retention in tumors due to being monovalent and thus lower avidity than a mAb. Intermediate sized fragments, especially divalent fragments such as diabody (~55 kDa), or minibody (~80 kDa) would provide rapid tissue penetration, high tumor retention and also rapid elimination (due to lack of FcRn binding) (39). Monovalent fragments such as Fab (~50 kDa) may not have the tumor retention while bivalent Fab<sub>2</sub> (~110 kDa) approaches the size of a full mAb while lacking the (potential) added benefit of the effector function. On the other hand, even smaller constructs such as DARPin (~14 kDa) or affibodies (~7 kDa) with very high affinities may be interesting. To our knowledge, none of the programs that are in the clinic currently use anything other than full-length antibodies.

### Conjugation Strategies: Site of Conjugation and Chemistries

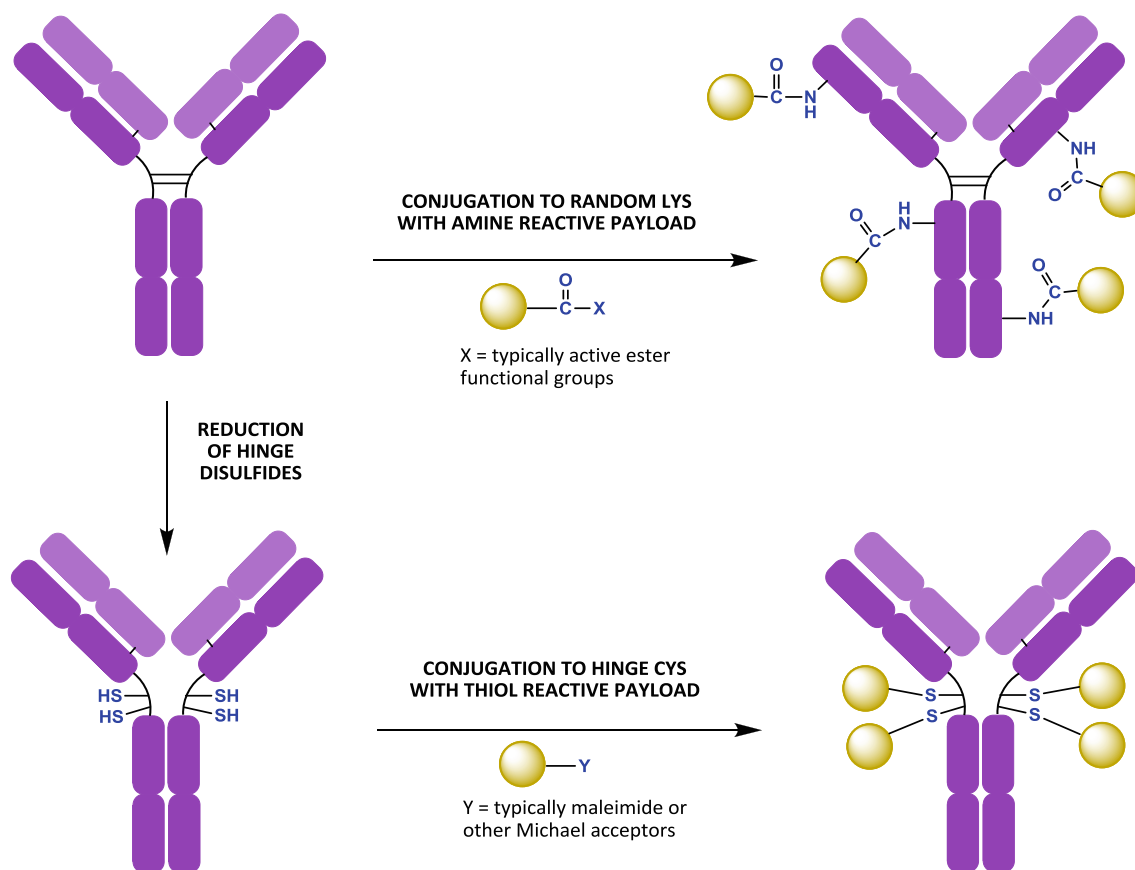
A key characteristic of an ADC is the amount (number) of drug molecules per antibody known as the DAR. While the theoretical DAR, dependent on the chemistry chosen, may be very diverse (for example 8 for cysteine-linked; much higher for lysine-linked), the actual DAR of an ADC is determined both by the chemistry chosen and the process employed to perform the conjugation. Conventional conjugation (Fig. 2) approaches where the conjugation occurs at solvent-exposed reactive amino acids (primarily cysteine and lysine) results in a heterogeneous population of species with varying DARs as well as variable conjugation sites. This diversity can result in challenges in maintaining product consistency, while the individual species may theoretically also have distinct PK/PD, safety and efficacy/potency properties. To overcome these drawbacks, site-specific conjugation approaches have been developed using antibody engineering. These engineered antibodies and the associated conjugation approaches are also discussed in the subsequent sections.

#### Conventional Conjugation

**Native Cysteine Conjugation.** Thiol coupling to reduced cysteines from the interchain disulfides can result in a theoretical limit of 8 payload molecules per Ab (for an IgG<sub>1</sub> or IgG<sub>2</sub>). Intrachain disulfides are less solvent exposed and more stable (40,41), and are generally not impacted by the reaction conditions used. However, complete coverage of the cysteines is

generally not desirable due to greater toxicity and increased clearance associated with the high DARs without a concomitant improvement of efficacy. DARs of 4 have been found to be optimal (42), although this conclusion could be target and payload dependent. There is also a stronger perturbation in structure and enhanced hydrophobicity of the high DAR molecules (43,44) reducing their stability and increasing the likelihood of formation of aggregates over time. Reports on impact of partial reduction of the disulfides on effector function are mixed (41), but a high DAR is more likely to have an adverse impact. Actual DARs in a pool of the product could range from 0 to 8 with each DAR (other than 0 and 8) composed of various isomeric forms, each with a theoretically unique set of properties. However, proper control of the process conditions can be used to ensure product consistency and reduce the heterogeneity obtained in the pool. Sun *et al.* (45) show that partial reduction by dithiothreitol (DTT) followed by conjugation with valine citrulline-monomethyl auristatin E (vc-MMAE) (see Table I for linker and payload) or full reduction by DTT followed by partial oxidation with 5,5'-dithio-bis-(2-nitrobenzoic acid) (DTNB) and then conjugation, led to DARs of 2, 4 and 6, but with 60–90% isomeric homogeneity within the process used. Brentuximab vedotin produced using above technology has an average reported DAR of ~4 (46). Non-denaturing hydrophobic interaction chromatography (HIC) can be used in purification with success and ease of elution dependent on hydrophobicity of linker-payload and the DAR. Highly hydrophobic linker-payloads along with high DARs may require high levels of organics for elution (personal communication).

**Lysine Conjugation.** Lysine-based conjugation has also been reported to lead to DARs of between 0 and 8 (16) although theoretically significantly higher numbers are possible due to the abundant surface lysines in mAbs. However, a high degree of conjugation would significantly alter the structure and properties of the mAb due to loss of positive charges associated with lysines, and thus a possible loss of antibody functionality. An average DAR of 3.3 has been reported for T-DM1 where surface lysines are functionalized with a bifunctional linker SMCC (containing *N*-hydroxysuccinimidyl ester for an amide bond with lysine and maleimide for attachment to payload) (47). Wakankar *et al.* (48) report a DAR of 3.5 for T-DM1 with actual substitutions varying between 1 and 7. In lysine-conjugated-huN901-DM1, at least 47% of the lysine residues (out of 86) were found to be partially modified. Almost all the modified lysines are to be found in regions of high flexibility and solvent accessibility (49). The low overall degree of conjugation and thus the distribution of substitution over a large number of lysines residues (7 in light chain (LC), 13 in heavy chain (HC)) leads to large degree of heterogeneity in the population of isomers but consequently also low deviation from the parent structure (20,49). There is strong likelihood that the



**Fig. 2** Conventional conjugation approaches for payload-linker attachment to antibody protein using (top) active ester coupling to random lysines or (bottom) Michael addition donors to reduced disulfide bridged cysteines in the hinge region of the antibody.

conjugation reaction conditions determine the sites of conjugation. The importance of the DAR and population heterogeneity is emphasized by the observation from pharmacokinetic studies that high DAR species in T-DM1 (DAR 3 to 7) deconjugate faster *in vivo* than DAR1 or DAR 2 (50). Proper process controls can reduce the heterogeneity of the population. In the case of gemtuzumab ozogamicin, the calicheamicin payload (Calich) is attached directly to lysines *via* an amide bond. An overall DAR of 2–3 is reported but the product contains approximately 50% unconjugated mAb molecules while the rest are loaded at DAR of 4–6 (51). The related compound CMC-544 (inotuzumab ozogamicin) has a DAR of 5–7 with less than 10% unconjugated mAbs (52).

### Site Specific Conjugation

The lysine and cysteine conjugation chemistries described above rely on non-specific electrophilic reactions with the resultant product profile controlled by process parameters. These strategies also results in a heterogeneous mixture with each subpopulation displaying different physicochemical and pharmacokinetic properties. The heterogeneity of these products and the presence of suboptimal molecules (fast clearing, higher toxicity and higher instability from the high DAR

species, potentially sub-potent low DAR species) have led to the development of strategies for site-specific conjugation. Such an approach potentially enables identifying and using the optimal locations of conjugation that impart the desired stability, potency as well as PK-PD properties to the ADC. An optimal DAR may also enable a balance between potency and toxicity, enabling a wider window between therapeutic and toxic doses, *i.e.*, a wider therapeutic index. A low DAR molecule will likely require a higher dose for efficacy but will also be better tolerated. Since the approach uses a design strategy to include the conjugation sites in the antibody, the heterogeneity of the resulting population is limited, including limiting the fraction of unconjugated antibody.

**Engineered Cysteine Conjugation.** McDonagh *et al.* (53) maintained the use of thiol coupling but reduced the heterogeneity and number of conjugation sites by systematically eliminating 1, 2 or 3 of the 4 interchain solvent accessible disulfides bonds by cysteine to serine mutations in a chimeric anti-CD30 IgG<sub>1</sub> antibody cAC10. The subsequent conjugation to MMAE (by maleimido-caproyl-vc-*para*-aminobenzyloxycarbonyl-MMAE, mc-vc-*pabc*-MMAE) resulted in near quantitative homogeneous substitution at the remaining cysteines. Literature reports (53) show that reduction of the interchain

**Table 1** Linker Abbreviations with Their Description and Corresponding Payloads

Abbreviation	Linker Name/Description	Linker Type	Payloads Used
AcBut	active ester-functionalized 4-(4'-acetylphenoxy) butanoic acid reacted with dimethyl hydrazide to form hydrazone linkage	Chemically labile (acid labile) linker	Calich
AcLys	acetyl-lysine linker	Reactive handle for transglutaminase enzymatic conjugation	Auristatins
AcPac	Active ester-functionalized (3-acetylphenyl)acetic acid reacted with dimethyl hydrazide to form a hydrazone linkage	Chemically labile (acid labile) linker	Calich
AEVB	hydrazone of 5-benzoylvaleric acid-AE ester	Chemically labile (acid labile) linker	Auristatins
C2-keto-Gal	2-acetonyl-2-deoxy-galactose	Reactive functional group enzymatically transferred to N-acetylglucosamine (G0)	Alexa Fluor 488, Biotin
DMA	dimethyl acid	Noncleavable linker	Calich
DMH	dimethylhydrazide	Linker precursor to hydrazone linkage	Calich
Gal NAz	N-acetyl-azido-galactosamine	Reactive functional group enzymatically transferred to N-acetylglucosamine (G0)	Alexa Fluor 488, Biotin
Phe-Lys	phenylalanine-lysine	Enzymatically cleavable linker	Auristatins
mc	maleimidocaproyl	Noncleavable linker	Auristatins
MCC	(N-maleimidomethyl)cyclohexane-1-carboxylate	Noncleavable monofunctional linker resulting from SMCC reaction with mAb	Maytansinoids
pABC	para-aminobenzylcarbonyl	Self-immolative spacer	Auristatins
pAcPhe	para-acetyl-phenylalanine	Unnatural amino acid used for selective conjugation with alkoxy-amine payloads	Auristatins
PEG	polyethylene glycol	Noncleavable spacer	Auristatins, Maytansinoids
SMCC	succinimidyl-4-(N-maleimidomethyl)cyclohexane-1-carboxylate	Noncleavable bifunctional reagent	Maytansinoids
SPDP	N-succinimidyl-3-(2-pyridyldithio)propionate	Disulfide-containing reducible linker	Maytansinoids

disulfides would not impact biosynthetic assembly of IgG, antigen affinity or PK properties. Removal of inter-heavy chain disulfides could impair effector functions. Replacement of the various cysteines with serine did not impact the antigen-binding. A reduction in efficacy and an increase in tolerability results in a net constant therapeutic index for DARs of 4 or 2. The site of conjugation was determined to be less critical for potency or tolerability compared to the stoichiometry, in agreement with Hamblett *et al.*, (42).

A variation of thiol-coupling after reduction of interchain disulfides was described in Klinguer-Hamour *et al.* (54). The technology involves reduction and subsequent relinkage using a bridging moiety that carries the payload. Conjugates formed with bridging moieties that included a polyethylene glycol (PEG, 11-mer) spacer, had reduced aggregation propensities compared to conventional conjugates. Miao *et al.* (55) also described an analogous technology called C-Lock™. Disulfide bridging technologies are specifically considered in a subsequent section on Linkers.

The first truly site-specific approach was developed by Junutula and coworkers (16,56) via the THIOMAB technology employing thiol coupling to engineered cysteines, without disruption of the interchain disulfides as in conventional cysteine conjugation. The THIOMAB technology involves selective introduction of unpaired cysteines at specific sites on the

surface of the IgG so as not to impact the folding (after expression), structure or functionality (antigen binding) of the mAb (namely THIOMAB). Conjugation to these cysteines then generates THIOMAB drug conjugates (TDCs) with homogeneous and narrowly defined drug loading. THIOMAB constructs (at HC-Ala114Cys Kabat; DAR of 2) retained efficacy but showed improved safety profiles with respect to bone marrow and liver toxicity, in various mAbs despite slower clearance ((16,56–58)). Lower effective hydrophobicity of TDCs (as well as other low DAR molecules with hydrophobicity arising primarily from the payload) may lower the hepatotoxicity, since the liver plays a key role in metabolizing hydrophobic macromolecules (59). Junutula *et al.* (2010) also observe that the mode of toxicity of drug conjugates was predominantly target antigen independent. Liver and bone marrow toxicities are attributable to the high rate of fluid-phase pinocytosis and unrestricted access to blood-borne antibodies in these organs ((59)). TDCs seem to be able to reduce the DAR while maintaining efficacy and thus improving hepatotoxicity. The overall benefit on bone marrow toxicity though is unclear.

The importance of the site selected for introduction of the cysteine residue (by replacing serine, valine, or alanine) in the THIOMAB has been illustrated by process and stability analyses. The LC-Val110Cys could not be as efficiently



conjugated as HC-Ala114Cys. Of 12 other sites (8 on LC and 4 on HC), 8 showed >90% conjugation efficiency (56), depending on solvent exposure (58). Similar DARs (1.7–1.9) were obtained when using THIOMAB constructs LC-Val205Cys, HC-Ala114Cys and HC(Fc)-Ser396Cys, and all constructs retained cell-binding and internalization properties of the native mAb (trastuzumab). However, *in vivo* efficacy and plasma stability studies with the TDCs showed that the LC-Val205Cys was the most active and stable, the Fc-Ser396Cys had little to no activity, while HC-Ala114Cys was moderate. Other Fc sites tested showed improved plasma stability suggesting that it was the site and not the domain that was critical.

The extent of solvent exposure of the introduced cysteine impacts not only its degree of substitution but also the *in vivo* stability when using the mc-vc-pabc linker. Loss of the conjugate occurs due to maleimide exchange with reactive thiols in plasma with albumin, free cysteine or glutathione (retro-Michael reaction) (58,60). However, not all sites are equally susceptible. Both HC-Ala114Cys and HC(Fc)-Ser396Cys are found in neutral charge environments but the extent of maleimide exchange was different, with HC(Fc)-Ser396Cys being the most susceptible. Conjugation site LC-Val205Cys on the other hand, is in a positively charged environment. In this case, the succinimide ring hydrolyzes rapidly and prevents the maleimide exchange to thiol-reactive plasma components from occurring (58). The retro-Michael reaction is clearly dependent on structural factors beyond solvent accessibility and local charge. Lower *in vivo* stability also correlated with increased liver toxicity, probably through uptake of albumin-drug conjugate, although it is not clear if these were dose limiting. Alley *et al.* (60) showed that attachment of drug to thiols using haloacetamide, by use of bromoacetamidocaproyl instead of the mc linker, circumvented the retro-Michael reaction, as discussed in detail later.

A modeling approach to help identify sites for introducing cysteine was proposed by Voynov, *et al.* (61). Fourteen IgG<sub>1</sub> cysteine variants were produced and analyzed for cross-linking or oligomerization propensity. The variants were designed so as to have at least one variant per immunoglobulin fold domain. The cross-linking propensity of the variants (of serine, threonine, asparagine or lysine) was related to the spatial-aggregation propensity (SAP) score of the substituted residue. [Calculated on a residue by residue basis, the SAP value of a particular region is a measure of the dynamic exposure of hydrophobic patches that is related to the aggregation propensity of that region (62)]. Sites with partial solvent exposure (SAP score 0.00 and –0.11) and away from hydrophobic patches (high SAP scores) are preferably modified with low impact on structure. A modification site with a high SAP patch in the vicinity is likely to result in structural abnormality (61). The model offers an interesting tool to rationally preselect sites for modification based on the consideration of risk for

oligomerization of the cys-variants. However, production of the ADC requires further processing steps, while the actual impact of the selected site on toxicity, PK-PD and efficacy, based on the current state of the knowledge, can only be discerned by (pre)-clinical testing.

**Enzymatic Conjugation.** Conjugation in the THIOMAB constructs requires a reduction step to first remove any cysteine or glutathione adducts from cellular sources on the engineered cysteine. This step is necessary due to the presence of the mutant unpaired cysteine. Any concomitantly disrupted inter-chain disulfides are then allowed to re-oxidize (under mild oxidation conditions such as air or by induced oxidation using CuSO<sub>4</sub> or dehydroascorbic acid), prior to the conjugation step (56). Triple light-chain antibodies were detected and addressed by adjusting cell culture conditions (63,64). Some of these processing issues have been avoided by an enzymatic method for site-specific conjugation (65). The microbial enzyme transglutaminase (mTG) catalyzes the formation of a covalent bond between a glutamine side chain and a primary amine. Although mTG shows some promiscuity in its recognition sequence, it does not seem to recognize naturally occurring glutamines in the constant regions of glycosylated antibodies (66). However, it catalyzes bond formation for a variety of structurally diverse payloads as long as a four-carbon linker (no side chains) separates the primary amine group (acyl acceptor) from the rest of the molecule (67). Strop *et al.* (65) therefore introduced a glutamine tag (with LLQGA amino acid sequence) at various locations to design specific conjugation sites and examine impact of the site on conjugation efficiency, stability, PK and toxicity. Exploration of about 90 sites in an anti-EGFR IgG<sub>1</sub> mab identified 12 sites with good biophysical properties and high degrees of conjugation. A variety of structurally diverse probes and payloads could be conjugated by using appropriate amine-containing linkers. Two conjugation sites were examined in detail (both at the C-terminus of the anti-M1S1 antibody C16; C16-LC-AcLys-vc-MMAD or C16-HC-AcLys-vc-MMAD) after conjugation with AcLys-vc-MMAD. DARs were found to be 1.8–1.9 in the range of the expected 2.0. Detailed characterization of these mTG-based ADCs has been published. Mass spectrometric methods (intact; tryptic and Glu-C digests) showed the site-specificity of this technology to be better than 95% (based on a detection limit of 5% for the DAR 2 species) thus resulting in highly homogeneous ADCs (68). No species with loading greater than 2 were detected. Higher DAR ADCs were created by adding the glutamine tags to C16 HC and C16 LC simultaneously, showing mostly DAR 4 with a few DAR 3 resulting in an overall DAR of 3.8. Interestingly, a small fraction (~1.3%) of conjugates at Q295 were detected, arising from the presence of a small fraction of aglycosylated antibodies in the pool. Lack of glycosylation at N297 allows the 293EEQYN297 sequence to also be conjugated by mTG. This variant could be

eliminated by mutation Q295N, yielding better than 99.8% specificity (68).

Potency and efficacy of the transglutaminase-based ADCs in a mouse model was in proportion to the ~2-fold drug load compared to a standard cysteine conjugated C16-mc-vc-MMAD with DAR of 3.6. The PK of C16-LC-AcLys-vc-MMAD and C16-HC-AcLys-vc-MMAD was similar in mice (for the first week), but the C16-HC-AcLys-vc-MMAD showed significantly faster clearance in rats (83% in first 24 h). Stability differences were seen in DAR 1.0 species between mice and rats. Proteolysis was found to occur at the vc-pabc linkage. The difference in clearance was not attributable to FcRn interactions, suggesting that heavy chain C-terminal mutations may influence other cellular interactions that affect clearance (69).

**Unnatural Amino Acids as Conjugation Sites.** Site-specific conjugation has also been achieved by introducing unnatural amino acids in specific locations to serve as the conjugation sites. Axup *et al.* (70) report on the incorporation of *p*-acetylphenylalanine (*p*AcPhe) in to an IgG<sub>1</sub> and then using the keto group of *p*AcPhe to selectively conjugate alkoxy-amine derivatized drug of interest through a non-cleavable oxime bond. This was illustrated by synthesizing anti-Her2 (Fab or IgG) with nAF (oxamine-auristatin F containing a short polyethylene glycol spacer). The *p*AcPhe was introduced at either position LC-K169X or LC-S202X or HC-A121X. Cell-based activity assays showed similar EC50s (1–10 nM) with LC-K169X being slightly lower. The conjugation kinetics are slow. Conjugation was performed in acetate buffer at pH 4.5 for 1–2 days at 37°C for the Fab or 4 days at 37°C for the IgG, with 20–30 fold molar excess of the AF. Detailed testing of the anti-Her2 IgG-nAF conjugate (HC-A121X) (DAR 2) showed no loss of activity after incubation in serum (3 days at 37°C), excellent potency in animal models, and PK similar to that of the naked mutant mAb. The impact of conjugation site on its PK was not studied. In a more recent and comprehensive report (71), the significant advantages of this site-specific technology was illustrated. *p*AcPhe was introduced into anti-5T4 and anti-Her2 antibodies in stable Chinese hamster ovary (CHO) cell lines with high titers (>1 g/L). ADCs were produced with both protease-cleavable and non-cleavable linkers. ADCs with non-cleavable linkers had improved *in vitro* and *in vivo* efficacy and pharmacokinetic stability in rodent models compared to cysteine-conjugated ADCs (both engineered cys and conventional cys). The improved *in vitro* efficacy and selectivity was attributed to stability of the linker before internalization and a more efficient intracellular drug release after internalization. For ADCs with cleavable linkers (vc-pabc), the site of insertion of the *p*AcPhe and therefore of the conjugation was found to significantly impact on stability in plasma. Single-dose rat toxicology study showed good tolerability of these ADCs up to 90 mg/kg (71). Behrens and Liu (69) judge the use of unnatural amino acids to be particularly well suited for scanning

potential sites of conjugation due to the simplicity of the incorporation technology. Clinical aspects of the technology and risk related to the unnatural amino acids and such bio-orthogonal linkages remain to be studied. Conjugation kinetics can be reasonably expected to be related to the position of the substitution. Sutro Biopharma, Inc. report that azido-based unnatural amino acids and dibenzocyclooctyl -based linker-payload have rapid kinetics with conjugation complete in a few hours on certain sites under room temperature and with only 5× molar excess of linker-payload (72).

**Glycan Conjugation.** Technologies have been proposed to use these glycans for site specific conjugation since the glycan attachment site is generally removed from antigen and FcRn binding site. It must be noted that while the glycans are not directly involved in the effector function interactions, the conformation of the essential IgG Fc interface is dependent on the presence of the oligosaccharide. Effector functions can therefore be severely compromised or lost in aglycosylated or de-glycosylated IgGs (33).

A glycan remodeling technology summarized in Klinguer-Hamour *et al.* (54) utilizes an enzyme to trim the glycans in an IgG followed by addition of a sugar carrying an azide, thiol or chloride moiety for further conjugation (73). Various technologies have been described for the addition of azide handle (74). Bicyclononyne attached to vc-MMAF was conjugated to the azide using copper-free click chemistry called strain-promoted alkyne-azide cycloaddition. The technology is proposed to be broadly applicable to all IgGs and can be extended by glycoengineering (*e.g.*, introducing new glycosylation sites or different sugar functionalities).

Another chemo-enzymatic approach is described by Boeggeman *et al.* (75) who use a mutant version of the enzyme galactosyltransferase to transfer a sugar residue with a reactive functional group (*e.g.*, C2-keto-Gal or Gal NAz) to *N*-acetylglucosamine. The N-linked (Asn297) glycoprotein is first enzymatically trimmed down to G0, purified, and then tagged with C2-keto-Gal from its UDP-derivative to the free GlcNAc residues on the mAb by the modified galactosyltransferase. Identification of the galactosidase enzyme for reproducible and complete trimming to G0 was critical. This remodeling of the glycans was shown by ELISA to not impact binding to antigen. A fluorescent probe was introduced using a C<sub>5</sub>-aminoxyacetamide linker. A conjugation level of 2 was achieved (equivalent to a DAR 2).

## Cytotoxic Payload

The second design aspect, the cytotoxic payload, is what makes the ADC a powerful therapeutic. The payload is the active species that kills a tumor cell typically through (1) interference with the tumor cell mitotic cycle by binding to tubulin, or (2) damaging DNA *via* cleavage or alkylation. Early work

with ADC payloads relied on clinically utilized drugs such as vinca alkaloids (4-deacetylvinblastine, tubulin inhibitor) (76), methotrexate, doxorubicin (77) and taxanes (paclitaxel, microtubule stabilizer) (78). However, development of these early ADCs was hampered by less than ideal payload potencies, heterogeneity and lack of specific targeted cytotoxicity. From a targeting standpoint, the fact that only a very small percentage of the injected dose gets localized to the tumor site requires that the payloads be highly potent.

The current payloads, either approved or in clinical development that are the focus of most ADC development, are all natural product-derived molecules that typically have picomolar or better potencies. The calicheamicins and duocarmycins are DNA binding cytotoxic agents while the auristatins (monomethyl auristatin E, MMAE; monomethyl auristatin F, MMAF) and maytansines (DM1, DM4) are tubulin inhibitors (79,80).

### Calicheamicins

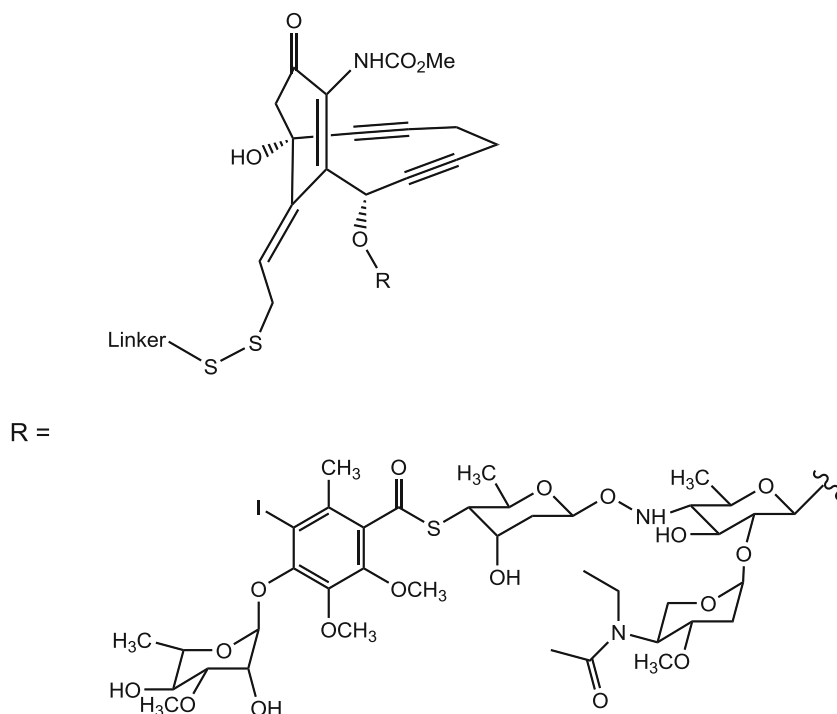
The calicheamicin family of antitumor antibiotics (81) are perhaps the most potent payloads with subpicomolar potencies. These cytotoxic semi-synthetic natural products were isolated from the soil microorganism *M. echinospora calichensis* and modified to incorporate a linker handle. The calicheamicin analog used in gemtuzumab ozogamicin is the *N*-acetyl- $\gamma$ -calicheamicin (Calich). In Fig. 3, the calicheamicin structure displays three general functional areas. The larger region includes four monosaccharide units and a hexasubstituted

benzene ring. This aryltetrasaccharide functionally binds tightly to the minor groove of double-helix DNA (82) with sequence specificity toward 5'-TCCT-3' and 5'-TTTT-3' (83). The next functional area contains a highly strained enediyne unit within a 10-member bicyclic ring. This rigid enediyne moiety is attached to an allylic disulfide trigger group. When the disulfide is cleaved under reducing conditions, the resulting nucleophilic thiolate can undergo a rapid 1,4-addition to give a dihydrothiophene. The strained enediyne will spontaneously undergo a cycloaromatization reaction generating a highly-reactive 1,4-benzenoid diradical that eventually leads to DNA double strand cleavage (83) and ultimately cell death. Finally, the third region on the opposing side of the disulfide bond is the linker functionality. Two types of linkers have been synthesized onto Calich, one containing a hydrazone cleavage site (AcBut) and one without (DMA). Both react with Lys on the monoclonal antibody through an *N*-succinimidyl ester functionality (84).

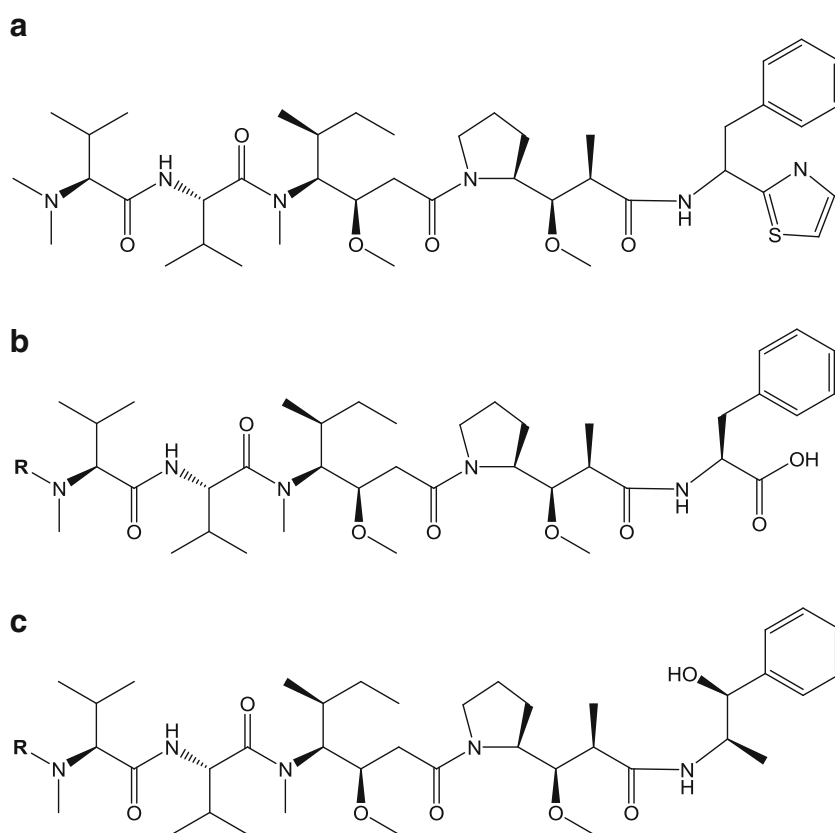
### Auristatins

The auristatin family of anti-mitotic cytotoxins were derived from the natural product dolastatin 10, a potent pentapeptide with unique amino acids (Fig. 4a), found in the Indian Ocean sea hare *Dolabella auricularia* (85,86). Many of the dolastatin analogs were investigated as anticancer, antimalarial (87) and antifungal (88) agents due to their ability to interact with the protein tubulin and inhibit its polymerization into microtubules thus causing cells in culture to accumulate in

**Fig. 3** Chemical structure of *N*-acetyl- $\gamma$ -calicheamicin payload (Calich).



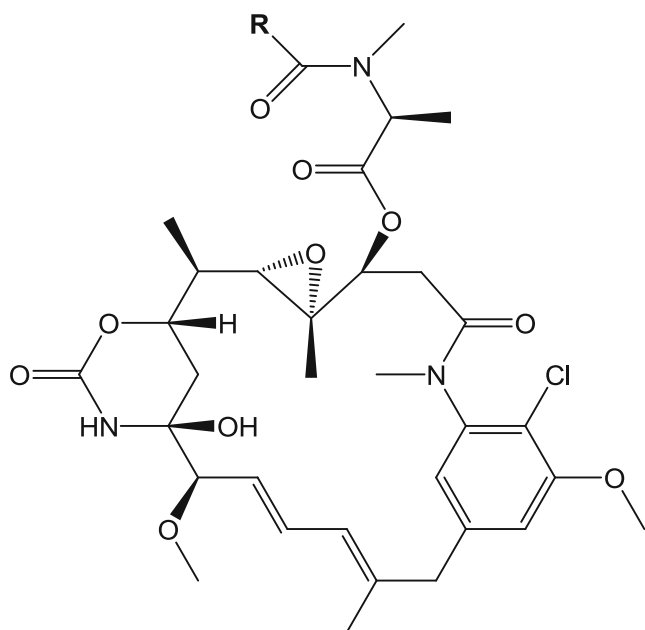
**Fig. 4** Chemical structures of auristatin payloads, **(a)** dolastatin 10, **(b)** monomethylauristatin F (MMAF) and **(c)** monomethylauristatin E (MMAE).



metaphase arrest. During studies to determine the mechanism of action on tubulin, it was found that the third amino acid (dolaisoleucine) was the most critical for interaction and that chiral centers were important for cytotoxicity (89). Since the total synthesis of dolastatin 10 was accomplished, multiple analog series have been synthesized. The auristatin series are closely related structural analogs of dolastatin 10 lacking the thiazole moiety on the C-terminal dolaphenine (Doe) residue. While a number of more polar amino acids at the C- or N-termini significantly decreased activity, the substitution of a phenylalanine for the Doe C-terminal residue coupled with the monomethyl-amine terminus yielded what is called MMAF (Fig. 4b) which retained good potency. MMAE (Fig. 4c) is analogous to MMAF but with a C-terminal 3-hydroxyphenethylamide (86,90). Seattle Genetics, Inc. has been pursuing the development of these auristatin analogs for use in their ADC technologies and brentuximab vedotin uses the MMAE payload. Multiple companies have licensed such technology from Seattle Genetics for use in their own in-house ADC projects and several ADCs using an auristatin are in clinical trials (91) [for example: Glematumumab vedotin (Celldex Therapeutics, Inc.), anti-CD79b (Genentech/Roche), anti-PSMA (Progenics Pharmaceuticals, Inc.), anti-AGS-16 (Agensys, Inc.), anti-GCC (Millenium/Takeda Co.), anti-5T4 (Pfizer, Inc.), and anti-EGFR (Abbvie, Inc.)].

### Maytansinoids

The maytansinoids are anti-mitotic tubulin inhibitors derived from maytansine, a benzoansamadolide antibiotic first isolated from the bark of the African shrub *Maytenus ovatus* (92) and subsequently found in the soil microorganism *Nocardia* sp (93). These cytotoxic agents bind to the same site on tubulin as the vinca alkaloids but are 100- to 1000-times more potent than clinically used anticancer drugs in killing cancer cell lines *in vitro* (94,95). Several Phase I and Phase II clinical trials were performed with maytansine but due to its poor therapeutic window, development of this agent was discontinued. Kupchan *et al.* (96) investigated the structural requirements for cytotoxic and antimetabolic activity with a library of structural analogs. They determined that the C-3 ester is necessary for activity but variations in the ester group are acceptable. Blockage of the C-9 carbinolamide *via* etherification reduces tubulin polymerization but does not strongly affect antitubulin activity (Fig. 5). Chari *et al.* (95,97) from ImmunoGen, Inc. were able to develop new, potent analogs of maytansine designed for conjugation to an antibody and release upon internalization within tumor cells. Two such analogs, N<sup>2'</sup>-deacetyl-N<sup>2'</sup>-(3-mercapto-1-oxopropyl)-maytansine (DM1) and N<sup>2'</sup>-deacetyl-N<sup>2'</sup>-(4-mercapto-4-methyl-1-oxopentyl)-maytansine (DM4), were synthesized and developed for further use as payloads on ADCs in the clinic. Ado-trastuzumab emtansine



DM1: R =  $-\text{CH}_2\text{CH}_2\text{SH}$   
 DM3: R =  $-\text{CH}_3\text{CH}_2\text{CH}(\text{CH}_3)\text{SH}$   
 DM4: R =  $-\text{CH}_2\text{CH}_2\text{C}(\text{CH}_3)_2\text{SH}$

**Fig. 5** Chemical structures of maytansinoid analog payloads,  $N^2$ -deacetyl- $N^2$ -(3-mercapto-1-oxopropyl)-maytansine (DM1),  $N^2$ -deacetyl- $N^2$ -(4-mercapto-1-oxopentyl)-maytansine (DM3) and  $N^2$ -deacetyl- $N^2$ -(4-mercapto-4-methyl-1-oxopentyl)-maytansine (DM4).

(T-DM1) uses the DM1 payload while the Phase II anti-CD19 conjugate, coltuximab ravtansine (previously called SAR3419), uses the DM4 payload (98).

### Duocarmycins

The class of duocarmycin analog payloads were based on the natural product duocarmycin isolated from *Streptomyces* sp. bacteria collected at the foot of Mt. Fuji, Japan in 1988 (99). These analogs bind to AT-rich regions of the minor groove of DNA with subsequent irreversible alkylation, disrupting the DNA and leading to cell death. This unique mode of action exerts its effect at any stage of the cell cycle, unlike tubulin binders which are only effective in the mitotic state of the cell (100). The reactivity begins when the duocarmycin binds to the DNA causing a conformational change to the otherwise stabilized cyclopropane moiety. Disruption of the vinylogous amide brings the cyclopropane into conjugation with the cyclohexadienone ring system and selectively activates it for adenine N3 alkylation of the DNA strand (101). This provides for the important benefit of multi-drug resistance, even for *P*-glycoprotein efflux pump expressing cell lines. These molecules, even though based on natural products, can be fully chemically synthesized and are thus not dependent upon isolation from natural sources. A novel prodrug analog of duocarmycin was

conjugated to a fully human anti-CD70 antibody to form the MDX-1203 ADC and used in a Phase I clinical trial (102,103).

Other natural product payloads investigated include pyrrolobenzodiazepines (PBDs, DNA crosslinking) (104), the cyclic octapeptide analogs of the mushroom amatoxins (RNA polymerase II inhibitor) (105), epothilones and anthracyclins. Additionally, there have been a few non-natural product-derived synthetic drugs investigated in preclinical studies but none have found much success, possibly due to their less than ideal potencies. Some future directions in payload discovery and development involve investigating more potent chemotypes and identifying novel tumor cell-killing mechanisms beyond those already being pursued (personal communication). Additionally, improvements to solubility and attenuation of hydrophobicity effect on parent antibody biophysical structure while not affecting binding activity or pharmacokinetics would be advantageous (8).

### Linker

The final design aspect of ADCs is the linker between the antibody and the toxic payload. This relatively short spacer moiety is a key component of an ADC structure. See linker reviews by (82,106,107). One important characteristic of an ideal ADC linker is that it should remain stable *in vivo* at neutral pH in the serum and also stay inert to any endogenous reactive molecules (*e.g.*, thiols, glutathione), enzymes (*e.g.*, cathepsins, proteases) or other biomolecules (albumin, immunoproteins). Additionally, the linker is required to release the toxic payload at the intended targeted site (such as within tumor cell lysosomal compartments). The stability of the linker and subsequent targeted release play important roles in the efficacy and safety/tolerability of ADCs. Premature release of the cytotoxic payload into the systemic circulation could prove damaging to normal tissue and result in clinically relevant side effects. Loss of conjugated drug on the tumor-targeting antibody also has a negative effect on potency. Therefore, choice of linker is an important consideration when designing an ADC.

The linkers are bifunctional, conjugating the relatively small molecule payload to the larger biomolecule antibody *via* orthogonal chemistries—one functional group to react only with the payload and another functional group to conjugate to the antibody. The ideal linker should involve simple, cost effective, high yielding chemistry for attachment to the toxic payload (typically organic solvent based) while providing gentle, aqueous-based bioconjugation to the antibody.

Most linkers fall into the broad categories of cleavable or non-cleavable. Cleavable linkers rely on specific proteases, pH environment within lysosomal compartment or reduction by endogenous thiols such as glutathione in the cytoplasm for their function for releasing. Non-cleavable linkers on the other hand, rely on proteolytic cleavage of the antibody itself in



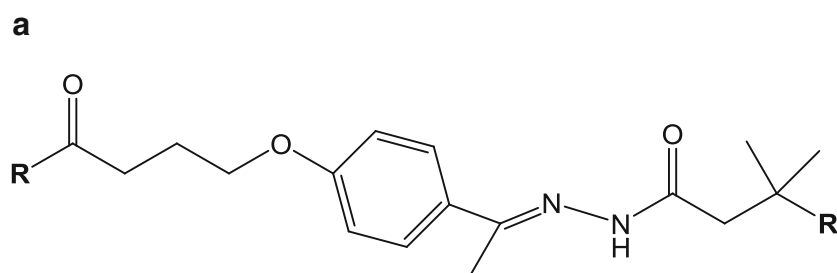
order to release the payload, which will still be attached to a portion of the linker and the amino acid residue at the site of conjugation.

Gemtuzumab ozogamicin and CMC-544 (a phase III ADC) use a cleavable linker to conjugate the toxic payload Calich to an anti-CD33 mAb for gemtuzumab ozogamicin (108) or to an anti-CD22 mAb for CMC-544 (52). The dimethyl hydrazide (DMH)-functionalized Calich payload, Calich-DMH, is reacted with the acetyl group of the active ester-functionalized 4-(4'-acetylphenoxy)butanoic acid moiety to form the AcBut linkage (Fig. 6). The active ester is used to conjugate to the lysine residues of the antibody with a resulting amide bond (106). During development of the gemtuzumab ozogamicin conjugate, it was shown that a hydrolytic cleavage site was necessary for potent and selective cytotoxicity from *in vitro* and *in vivo* rodent studies - this in spite of the presence of a reducible disulfide bond on the calicheamicin structure (84). This disulfide is stabilized by two adjacent methyl groups which likely prevent premature release of the payload due to endogenous circulating thiols mentioned above. It was found that stabilizing this disulfide bond improves the therapeutic index of the resulting ADC. Hinman *et al.* (81), using a number of linker analogs, found that the serum stability of the disulfide was proportional to the steric bulk placed adjacent to the disulfide and that the two methyl groups provided adequate balance between stability in circulation and release in lysosomes. To modulate (109) the stability of the hydrazone moiety, a broad variety of aromatic aldehyde and ketone precursors were used to make hydrazone linkers and were screened in an *in vitro* assay (108). In this assay, the hydrazone

linker conjugates were incubated for 24 h at pH 4.5 and pH 7.4 at 37°C to approximate the pH in lysosomes and in circulation, respectively. The hydrazone linker made from the AcBut was hydrolyzed at 97% at pH 4.5 but only 6% at pH 7.4. This stability difference plus *in vitro* cytotoxicity, selectivity and *ex vivo* testing were the reasons this AcBut linker was chosen for gemtuzumab ozogamicin. Similarly, for CMC-544, a number of linkers were investigated including acid-labile linkers (AcBut and (3-acetylphenyl)acetic acid (AcPac)) with Calich-DMH and an acid-stable amide linker, *N*-acetyl  $\gamma$ -calicheamicin dimethyl acid (Calich-DMA) (110) (Fig. 6). The AcBut and AcPac hydrazone linkers hydrolyze at similar rates under acidic conditions while the amide linker is not expected to hydrolyze significantly under those same conditions. Any internalized amide-linked conjugate would likely require more extensive proteolytic degradation than the "quick" release of the acid-labile linkers. This is supported by the fact that the acid-labile linked AcPac conjugate was 4- to 8-fold more potent than the amide-linked conjugate in CD22<sup>+</sup> B-cell lymphoma (BCL) cytotoxicity assays and that the AcBut linker conjugate was the most efficacious in a BCL tumor-mouse model. The AcBut linked conjugate was chosen for further development (CMC-544) as it caused greater and longer lasting inhibition of Ramos B-cell lymphoma growth than AcPac linked conjugate (85% vs 28% of mice being tumor-free after 50 days) (110).

Seattle Genetics brentuximab vedotin provides another example of a cleavable linker system on an ADC. In order to improve plasma stability while still affording drug release after tumor cell internalization, a protease-cleavable peptide linker

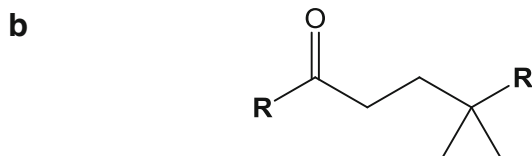
**Fig. 6** Chemical structures of (a) 4-(4'-acetylphenoxy) butanoic acid (AcBut) and (b) dimethyl acid linker (DMA) linkers.



AcBut (Hydrazone) Linker

R = N-hydroxysuccinimidyl ester or -OH

R' = disulfide-linkage to N-Ac-Calicheamicin



DMA Linker

R = N-hydroxysuccinimidyl ester or -OH

R' = disulfide-linkage to N-Ac-Calicheamicin

was developed for use with the MMAE payload (46). This vc linker is composed of the dipeptidyl substrate, valine-citrulline, linked through the amine terminus to a self-immolative spacer *p*-aminobenzoyl carbonyl (*pabc*) (Fig. 7). The vc linkage is stable in plasma but is very rapidly hydrolyzed by lysosomal enzymes, in particular, Cathepsin B. Upon cleavage of the vc, the *pabc* group undergoes spontaneous [1, 6]-fragmentation and releases the MMAE payload in native form (90). Initial studies compared the vc linker with an acid-labile hydrazone of 5-benzoylvaleric acid-AE ester (AEVB). Antibody conjugates of maleimido-vc-MMAE and maleimido-AEBV were made using a chimeric mAb cBR96 and *in vitro* plasma stabilities studies (mouse and human plasma) were performed. The hydrazone-linked AEBV was released from human plasma with a half-life of 2.6 days while the vc-linked MMAE was much more stable with a projected half-life of 230 days. While both conjugates were potent by *in vitro* cytotoxic assay, the mAb-AEBV conjugate was more toxic *in vivo* with a maximum tolerated dose (MTD) of 15 mg/kg compared to 30 mg/kg for the mAb-vc-MMAE conjugate.

Another example of a cleavable linker system is the disulfide based linker. Scientists at ImmunoGen (111), working with the maytansinoid payloads, set out to identify disulfide based cleavable linkers that would leverage the 1000-fold difference in endogenous thiols (*i.e.*, glutathione) from human circulating blood (typically micromolar range) to the cytoplasm of cells (typically millimolar range) (95). Several conjugates were made with huC242 antibody, a humanized IgG<sub>1</sub> that binds CanAg, a tumor specific antigen, varying the degree of steric hindrance (0, 1, or 2 methyl groups) on the geminal carbon to the disulfide bond on the linker side and on the maytansinoid side of the disulfide bond (111). The rates of reduction of the disulfide linkages in conjugates were measured at 37°C, pH 6.5, *in vitro* and it was found that the steric hindrance on the linker side produced greater increases in stability than steric hindrance on the maytansinoid side while

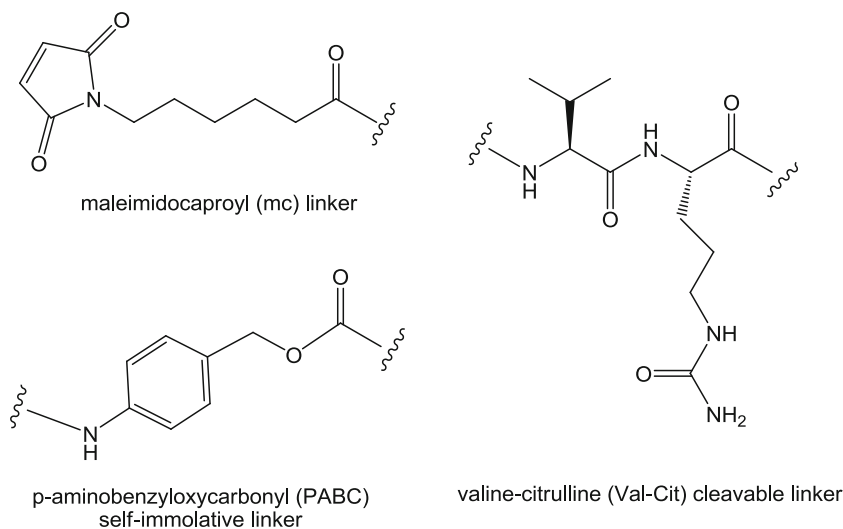
methyl groups on both sides of the disulfide produced the greatest stability toward reduction. Based on CD-1 mice pharmacokinetic studies, the rate of loss of maytansinoid from conjugate during circulation in plasma correlates with the level of steric hindrance at the disulfide bond which suggests that release of payload is likely occurring through reductive cleavage. The corollary to this is that the rate of conjugate clearance from plasma increases as the stability of the disulfide linker increases. *In vivo* efficacy studies using HT29 and COLO205 tumor models, however, indicate that there are more factors than linker stability involved with *in vivo* efficacy.

An example of an ADC with a non-cleavable linker is T-DM1. The antibody is conjugated at lysines *via* the succinimidyl ester reactive group on the bifunctional SMCC linker and the DM1 payload is attached through its thiol moiety to the maleimide functionality on SMCC. Interestingly, when comparing this non-cleavable linker to disulfide linkers (reducible/cleavable linkers), the serum concentrations of T-DM1 retained elevated concentrations compared to those ADCs with reducible linkers. Additionally, in a target independent study in rats, toxicity was lower for the T-DM1 with SMCC than a reducible SPP ADC comparator (112). It was hypothesized that in order to release the payload from the non-cleavable linker, that portions of the antibody would need to be catabolized resulting in amino acid-capped-linker-payload. Shen *et al.* (113) performed a preclinical study that identified the major catabolites of T-DM1 to include DM1, [N-maleimidomethyl] cyclohexane-1-carboxylate-DM1 (MCC-DM1), and Lys-MCC-DM1. These species of ADC linker-payload catabolites were also detected in patient plasma samples.

### Next Generation Linker Technologies

Ambrex, Inc. has proprietary bacterial (ReCODE) or yeast/mammalian (EuCODE) expression systems which incorporate

**Fig. 7** Chemical structures of non-cleavable maleimidocaproyl (mc) linker, *p*-aminobenzyl-oxycarbonyl (PABC) self-immolative linker and cleavable valine-citrulline (vc) linker.



non-native amino acids (nnAA), such as *p*-acetylphenylalanine (*p*AcPhe), into expressed proteins through the ribosomal use of an Ambrix-engineered tRNA synthetase. These site specific incorporated nnAAs react bioorthogonally with custom hydroxylamine linkers to yield site-specific antibody-drug conjugates (NDCs) (109). In two model systems using *p*AcF engineered anti-5T4 and anti-HER2 antibodies, hydroxylamine-functionalized monomethyl auristatin D, with either cleavable or noncleavable linkers, were conjugated site-selectively to the incorporated nnAAs of the mAbs. The NDCs using non-cleavable linkers proved more stable and displayed improved efficacy over the conventional ADCs with maleimide linkers attached at cysteines resulting from interchain disulfide reduction or cysteines that were engineered at surface-exposed positions (70,71). Though promising, the full potential of these NDCs have yet to be determined in clinical trials.

Allozyne, Inc. is another company focusing on site-specific engineering of nnAAs (114). Their Biociphering™ platform originated in the labs of David Tirrell at Cal Tech (115). This technology focuses on the incorporation of azido- or alkynyl-analogs of certain amino acids onto tumor-targeting mAbs and conjugation of the resulting site-specifically incorporated reactive groups to custom bioorthogonal payload linkers using Staudinger Ligation (or “click” chemistry) to create “AzAbs”. The click reaction involves the copper catalyzed cycloaddition of an azide with an alkyne to produce a triazole. It is claimed that the aromatically stabilized linkage is more stable than maleimide, thioester or oxime chemistries. However, published data is sparse. Allozyne is pursuing undisclosed ADC targets for oncology (116).

Redwood Bioscience (now part of Catalent, Inc.) is developing SMARTag™ technology which utilizes site-selective aldehyde placement and aldehyde specific chemistry to control site of conjugation and payload stoichiometry (117). This bioconjugation chemistry called “aldehyde tagging” was initially designed by the Carolyn Bertozzi group from UC Berkeley (118). A formylglycine-generating enzyme (FGE) recognition sequence consisting of five amino acids (CxPxR) is inserted specifically at intended conjugation sites. The FGE selectively alters the cysteine residue in the sequence to an aldehyde bearing formylglycine residue (fGly). This formyl moiety reacts with aldehyde specific payload linker reagents. One recent report identifies a hydrazine-iso-Pictet-Spengler (HIPS) ligation of a fluorophore to an anti-HER2 antibody as model system for ADCs. The HIPS ligation reagent was found to be superior to aminooxy compounds due to faster reaction kinetics and improved hydrolytic stability of the resulting oxacarboline linkage in buffer or serum stability (119). Agarwal *et al.* (120) claimed that in an *in vivo* efficacy study of SMARTag™ HER2 ADC, no free drug was observed over 28 days. However, no analytical details on limits of detection were revealed. Drake *et al.* (121) studied a panel of

ADCs with the aldehyde tags at different locations on the antibody (trastuzumab) selected to minimize perturbation of native structure but accessible to conjugation. There was no impact on thermal stability compared to unconjugated antibody and only a slight reduction on FcRn off-rate. When compared to a T-DM1-like lysine conjugated ADC, the aldehyde tagged ADCs showed improved PK and lesser toxicity in rodents.

Disulfide bridging linker technologies from Polytherics Ltd, Thiologics, Ltd. or Igenica, Inc. are another approach at site specific conjugation which occurs at the disulfide bridges between light and heavy chains. This category of linkers include Polytherics Ltd.’s ThioBridge™ technology, Thiologics, Ltd.’s 3,4-substituted maleimide linkers and Igenica, Inc.’s SNAP technology. Polytherics’ Thiobridge™ linker is composed of a thiol-specific bis-sulfone moiety that is capable of sequential addition-elimination across a disulfide bond (122). In the case of an ADC (123), this would entail the interchain disulfides of the hinge region (heavy-heavy) or between the light and heavy chains. The resulting conjugate contains a three-carbon bridge at the site of the reduced disulfide bond. This presumably would keep the interchain stabilized as well as provide attachment for the toxic payload. Thiologics’ 3,4-substituted maleimide linkers work in a similar fashion yet with different reactive chemistries that yield a rigid two-carbon bridge without the introduction of new chiral centers (124). Dihalo- or dithiophenolic-maleimide linkers have been used *in situ* to attach PEG polymers across the disulfide bond of somatostatin (125) or across the disulfide of an antibody fragment with nitroxide spin labels for diagnostic continuous-wave electron paramagnetic resonance or “spinostics” (126). Castaneda *et al.* (124) used this rigid linker to improve interchain stability for a doxorubicin-Her2 Fab conjugate. Igenica, Inc.’s SNAP ADC technology (previously called Stapled ADC) provides payload attachment at each of the disulfide bonds in a mAb (127,128). To date, however, none of these 3 disulfide bridging linker technologies have been fully developed for clinical use.

## ANALYTICAL METHODS FOR PHYSICOCHEMICAL STABILITY

Critical to the efficacy and safety of the ADC is stability of the linker, the chemical stability and potency of the drug species (once it has been conjugated), and the DAR. Approaches and methods that have been used for analytical characterization of ADCs have been reviewed in detail by Wakankar *et al.*, (129). Some of the key analytical methods, focusing on stability indicating assays for formulation development purposes, are briefly outlined here along with some examples of their application.

## Drug Antibody Ratio (DAR)

The drug antibody ratio of an ADC can be monitored by a number of methods. The simplest and most predominant method is the spectroscopic analysis by UV/Vis of the ADC (129). This method only requires that the UV/Vis spectrum of the antibody and the payload have different maximum absorbance wavelengths in order to calculate the concentration of both and thereby derive the DAR. A number of different ADCs have applied this technique such as calicheamicin analogs (81), immunoconjugates of doxorubicin (130), monomethyl auristatin conjugates (42) and maytansinoid conjugates (97).

Hydrophobic interaction chromatography (HIC) can also directly measure the DAR as well as drug loading distribution. HIC separates individual species based upon the degree of conjugation and then employs UV absorbance for detection. This separation not only allows for direct determination of the various conjugated species present but also allows for isolation and purification of the various species, if required for detailed characterization. (The possible utility for use of HIC in purification has been discussed earlier). The DAR is calculated based on the weighted average of the different species. The HIC method is not as easily applied to ADCs obtained from lysine conjugation since the higher heterogeneity of substitution (range of drug loading and sites), complicates the chromatographic separation (131). HIC is best applied to cysteine or site-specific conjugates where distinct well-defined species are present. In the case of MMAE conjugated to the anti-CD30 mAb, peaks corresponding to two, four and six DAR were identified (42). To assess the impact of high drug payloads and ionic strength to the overall stability of auristatin ADCs, the HIC method was used to separate fractions and the various DAR species were then collected and incubated (132). After incubation in low and high ionic strength buffers, aggregate content was measured by size exclusion chromatography (SEC) and it was determined that the aggregate content in the presence of high ionic strength was higher for certain DAR species (DAR 2 and DAR 4).

## Biophysical Analysis

Despite the size difference between a four-chain antibody and the cytotoxic drugs these small hydrophobic payloads can have a significant effect on the ADCs stability, solubility and solution properties (129). Biophysical techniques commonly used for monoclonal antibodies can also be used to assess the impact of conjugation, but the presence of the bound drug may complicate the characterization in certain cases.

It has been shown that the folding and assembly of the antibody is not affected by the conjugation process (56,133). This is likely to be process dependent but more often, attaching small hydrophobic payloads to an antibody has

been shown to cause subtle changes in secondary and tertiary structures. These changes can be studied by spectroscopic techniques such as circular dichroism (CD), intrinsic fluorescence, second-derivative UV spectroscopy and Fourier transform infrared spectroscopy (FTIR). However, many of the cytotoxic drugs used in ADCs exhibit a strong near-UV absorbance which can interfere with the characterization of the tertiary structure when using near-UV CD. Far-UV CD and FTIR on the other hand can be used to study the effect of the conjugated drug(s) on the overall secondary structure of the mAb (43,44).

Thermal analysis methods can also be used to study the impact of the conjugation on the conformational stability of the antibody. The most common is differential scanning calorimetry (DSC) (43,44,48,132). Differential scanning fluorimetry has also been used for studying antibodies but to date no reports have been published with the use of this technique with ADCs. It is possible that the dye used as the reporting fluorophore could interact with the hydrophobic payload, confounding the signal.

## Particle Characterization

The conjugation of hydrophobic payloads to a mAb can lead to aggregate formation by way of increasing the hydrophobic nature of the mAb. The conjugation of drug can also disrupt local secondary and tertiary structure resulting in unfavorable conformational changes. Just as with other protein products, the presence of aggregates and/or particulates is detrimental to the formulation development of ADCs. The standard tools used to examine particulates for unconjugated antibodies are also applicable for ADCs, and are reviewed in (134–136). To our knowledge, there are no specific reports of particulates in ADC products. Particulates assessment and characterization would be required for the ADC products as well as of their diluted infusion solutions, as discussed above [see also USP<1787>; (137)].

## Analysis of Residual or Dissociated Free Drug in Drug Product

An important aspect of the ADC physicochemical stability is the measure of free drug or unconjugated drug that has either dissociated from the antibody upon storage or is a residual amount that remained in the final drug product. The quantity of free drug poses concerns for both differential toxicity and safety.

The unconjugated drugs are typically small and hydrophobic which makes separation and quantification by reverse-phase high performance liquid chromatography (RP-HPLC) possible. For example, RP-HPLC was used for quantitative determination of free methotrexate (MTX) in the MTX-antibody conjugates on stability (138). It was demonstrated

that no detectable amount of free MTX in hydroxylamine-treated conjugates was present on stability in contrast to the presence of increasing amounts of MTX in samples produced at lower pH values. In an evaluation of an acid-sensitive linker, RP-HPLC was used to monitor the release of adriamycin from monoclonal antibodies in mildly acidic conditions and it was found to be rapidly released by hydrolysis in a pH range of 4 to 5.5, conditions typically found in endosomal and lysosomal vesicles (139).

Free drug may also be evaluated by competitive enzyme-linked immunosorbent assays (ELISA). For example, Tolcher *et al.* (140) determined the amount of non-protein-bound maytansinoid in a protein free extract of patient plasma. The free maytansinoid concentrations of two patients (dosed at the highest ADC levels in this study) could be detected in the plasma up to 48 h following treatment, 52 and 56 nmol/L immediately following cantuzumab mertansine infusion and declined to 4.6 and 5.5 nmol/L by 48 h. Based on this data, the free drug at all time points was less than 1% of the total circulating drug (free drug plus antibody-conjugated drug). To quantify the amount of free calicheamicin in gemtuzumab ozogamicin, Buckwalter *et al.* (141), employed a competitive ELISA method that used specific rabbit anti-calicheamicin antibody. Levels of unconjugated anti-CD33 antibody, total calicheamicin derivatives, and unconjugated calicheamicin derivatives were then used to characterize the pharmacokinetics of the drug.

### In Vitro Activity

An *in vitro* cell assay is used to evaluate the potency of ADCs upon release and storage as a quality parameter. There are various cell viability and apoptosis assays that can be performed *in vitro* to monitor the effects of the antibody-drug conjugate on the targeted cell. One example is a luminescent cell viability assay that can determine the number of viable cells in a culture after exposure to the ADC of interest based on the quantitation of the ATP present, an indication of metabolically active cells (142). One can also evaluate the effects of the ADC of interest on cellular health by a fluorescence-based method. The method uses a fluorescent dye that reacts to the reducing capability of living cells thereby allowing one to generate a quantitative measure of viability and cytotoxicity (143,144).

## PHYSICOCHEMICAL STABILITY

Formulation approaches and challenges for an ADC are similar to that of a conventional antibody or antibody fragment. Just as with antibody therapeutics there is a need to improve the physicochemical stability of an ADC. Therefore there is a need to understand how the ADC and the parent

antibody differ in terms of physical and chemical stability and how to go about formulating both molecules. Various aspects of the approach are discussed in the next section.

### Physical Stability

Conjugation in the ADCs involves modification of residues and the addition of bulky hydrophobic payloads. Impact on structure and stability is therefore expected. Higher order structural analysis shows perturbation but no significant global change for cys and lys conjugations. As discussed in a previous section, the greater the degree of substitution, the greater the destabilization of the ADC.

Wakankar *et al.* (48) showed that the melting temperature of the trastuzumab antibody CH2 domain decreased when only the linker was attached (T-MCC) and then decreased further upon conjugation to the drug DM1 to form (lysine-conjugated) T-DM1. It was also noted that there was reversibility of the first transition for the trastuzumab antibody and the conjugated antibody (T-DM1) but, in the case of the linker intermediate T-MCC, it was only observed to be partially reversible. The greatest impact of conjugation was on the CH2 domain which agrees with the analysis that the lysines in the CH2 domain are associated with greater flexibility and are therefore also more likely to be conjugated (48). Although lysine conjugation removes a positive charge from the mAb (also see below), the “light smear” of substitution is not expected to perturb the structure significantly (49). In the case of T-DM1, T-MCC and trastuzumab, the thermal stability did not completely correlate to aggregation behavior. T-MCC had highest rates of aggregation under thermal stress (at 40°C), followed by T-DM1 while trastuzumab had very negligible rates—which also explains the lack of reversibility of the first transition in T-MCC in the DSC analysis. Unlike trastuzumab or T-DM1, the majority of T-MCC aggregation was driven by the unconjugated maleimidyl moiety reacting with side chains of nucleophilic amino acid residues (cysteine, serine or tyrosine) on other molecules leading to formation of covalent aggregates. More significantly, the T-DM1 also showed higher rates of thermal aggregation than the trastuzumab which is in line with the expectation of destabilization on conjugation.

Qualitatively similar results were reported for the cysteine conjugates by Beckley *et al.* and Guo *et al.* (43,44). The conjugated mAbs were less stable to thermal aggregation compared to the parent mAb. Conjugation after reduction of disulfide bonds did not measurably perturb the secondary and tertiary structure, but did reduce the stability of the structure as seen by reduction of melting points, loss of reversibility of CH2 melt, lower energetics of melting and less cooperative melting. Again, the CH2 domain was more significantly impacted compared to the CH3 and Fab. Two regions in the Fc (in the CH2 and in the CH2-CH3 interface) were found to have



greater conformational dynamics in the ADC (cysteine conjugated vc-MMAE/mc-MMAF) compared to the parent mAb by hydrogen-deuterium exchange MS (145), providing orthogonal confirmation of the DSC results. Interestingly, the increase in conformational dynamics was caused by the breakage of the interchain disulfide bonds than by the conjugation (145). On the other hand, molecular modeling calculations suggest a substantial decrease in the conformational energy of the mAb upon conjugation—more so compared to impact of breakage of disulfide bonds only (44). The local surface around the conjugation sites also becomes more hydrophobic in the ADC, reducing the colloidal stability. The combination of lower conformational stability as well as lower colloidal stability compared to the mAb, results in a greater tendency of the ADC to aggregate (44). This is in agreement with the findings of Beckley *et al.* (43) and Adem *et al.* (132) who also showed that higher DAR species are more prone to aggregation due to the greater extent of destabilization (more disulfides broken) and even lower colloidal stability. The higher DAR species, due to their greater hydrophobicity, also enhance the risk for interfacial aggregation (44). Adem *et al.* (132) also report higher fragmentation rates for high DAR (6,8) species, which is related to the greater loss of interchain disulfide bonds.

Conjugation to lysines has less impact on thermostability than to cysteines as may be expected by the nature of the perturbation involved. Two different antibodies using the same chemistry and resultant DARs may also have different extents of destabilization. The extent of destabilization of an antibody for a specific type of conjugation (*e.g.*, lysine amine or cysteine thiol) is also dependent on the linker (for the same DAR) (146); however, the impact of linker could be overwhelmed by that of the payload for large hydrophobic payloads. Acchione *et al.* (146) also examined impact of coupling to sugar residues on the N-linked glycans using PEG (4-mer) and hydrazide linker. These conjugates had the least impact on thermal stability.

The impact of lysine conjugation on charge profile of an ADC is more complex than a simple loss of a positive charge for one payload added. Even a DAR of 1 can result in multiple charged isoforms depending upon the site of conjugation through impact on local (structural) environment, potential loss of salt bridges, perturbation of electrostatic network and influence on pKa. Higher DARs could also arise from conjugation of partially buried lysines which can alter the conformation and expose buried charged residues. Thus a greater heterogeneity in the electrostatic properties of conjugates is observed compared to that expected solely on the basis of lysines residues modified (147). Maeda *et al.* (148) also found a large number (>20) of peaks in the electropherogram (by capillary isoelectric focusing) for an IgG<sub>4</sub> ADC compared to that of an (unrelated) IgG<sub>4</sub> mAb ((8) peaks). Since change in charge (shifts in pI between 0.5 and 1 units) can alter PK and

tissue distribution of a protein (35), an understanding of the charge heterogeneity after conjugation is important. Proper control of the process to limit the range of sites will reduce the range of charge heterogeneity seen.

In summary, ADCs show reduced conformational stability compared to their parent mAb. They tend to be more hydrophobic due to the payload, thus reducing their colloidal stability also. This effect is proportional to the DAR, and is broadly independent of the chemistry involved. Structural perturbation is mainly limited to the domains that carry the conjugation with distant domains being less impacted. Lysine conjugation reduces net charge but also creates a heterogeneous mixture of a variety of charged isoforms. All these properties must be considered when designing the formulation for the ADC.

## Chemical Stability

### Chemical Stability of the Antibody

ADCs can inherit the chemical liabilities of their parent mAb. Protein chemical liabilities have been well studied and understood (149), and will not be discussed here. It is clearly important that the chemical modifications of the mAb as a consequence of chemical liabilities, do not impact the binding properties and activity of the mAb/ADC. Chemical hotspots are preferably addressed in the mAb design/selection phase itself, to prevent their carry-over into the ADC. There exists the possibility that the structural and conformational impact of conjugation results in exposure of buried (in the mAb) chemical hotspots making them active. However, as discussed above, the impact on structure and conformation is small so the risk of this effect is also small.

### Chemical Stability of the Payloads and Linkers

The chemical stability of the conjugated linker-payload affects the conjugation, purification, formulation and analytical development of an ADC. Conjugation and purification requires conditions that do not affect the molecular structure of the payload or linker. Formulation requires the stabilization of physical and chemical degradation pathways for both the antibody protein and small molecule linker-payload for long-term shelf storage. Analytical methods require sample preparation, matrices and solvent systems that have the potential to affect the sample being tested and, thus an understanding of the chemical stability is crucial (129). There is much in the literature regarding chemical stability of monoclonal antibody therapeutics (149–151). This section will focus on the chemical stability of the linker-payloads under *in vitro* conditions.

The chemical properties of the linker-payload as well as the site of conjugation on the antibody are major determinants in the chemical stability of the linker-payload (129). Changes

that affect the molecular structure of the payload and/or integrity of the linker attachment to the antibody potentially could lead to increased toxicity or decreased efficacy and could effectively change the therapeutic window of the ADC. Preserving the intact conjugate and preventing the release of free toxic payload during shelf storage is of critical importance in the formulation and development of the ADC drug product.

**Calicheamicins.** The calicheamicin payload-linkers contain hydrazone and disulfide moieties that are susceptible to *in vitro* chemical degradation under certain conditions. These ADCs have been characterized with several different analytical techniques including UV, IR, HPLC, and matrix-assisted laser desorption ionization-mass spectrometry (MALDI-MS) and electrospray ionization mass spectrometry (ESI-MS) (152). The hydrazone linker, found in gemtuzumab ozogamicin which is designed to hydrolytically release at lower pH in the lysosome, is acid-labile and known to hydrolyze in the presence of succinic acid during MALDI-MS or acetic acid in the ionizing solvent in ESI-MS during analytical testing (152). Calich is also susceptible to UV photolysis whether as the free drug or as a payload conjugated on an antibody. Low intensity UV irradiation of the free derivative Calich-DMA afforded no parent molecular ions (theoretical  $M+Na^+m/z$  1500) but ions that corresponded to some degradation product ( $M+Na^+m/z$  1322, likely a related sodium ion adduct of structure with MW 1299). This degradation product is consistent with loss of an  $H_2S$  group although no structure has been assigned. When Calich-DMA-hCTM01 mAb conjugate was irradiated with UV, the mass centroid of the broad MALDI-MS peak decreased in MW with increasing UV exposure. Presumably, the Calich-DMA payload undergoes UV photolysis leading to loss of payload from the antibody.

**Auristatins.** Senter *et al.* have developed the use of auristatin payload-linkers for use in ADCs and examined chemical stability as one criteria in their designs (60,90,153). Two auristatin payloads, AE and MMAE, were found to be stable for 16 days in buffered saline at 37°C. Additionally, *in vitro* liver lysosomal extracts or proteases did not appear to metabolize or degrade the auristatin payload molecular structures (90). In the course of designing their linker-payloads, they compared the pH-sensitive linkers against the protease-cleavable linkers. AE was synthesized with a pH-sensitive hydrazone linker at the C-terminus of the pentapeptide by condensing a maleimidocaproyl hydrazide with an AE ketoester. The resulting hydrazone of 5-benzoylvaleric acid AE ester (AEVB) was relatively stable at pH 7.2 ( $t_{1/2} > 60$  h) but was labile at pH 5.0 ( $t_{1/2}$  3 h). The AEVB was conjugated *via* the maleimide functionality to the cBR96 mAb and was also found to be stable at pH 7.2 ( $t_{1/2}$  183 h) and released payload at pH 5.0 ( $t_{1/2}$  4.4 h). In comparison, the protease-

cleavable dipeptide linker was attached at the N-terminus of MMAE through the self-immolative *pabc* spacer to form vc-*pabc*-MMAE. Since peptide bonds are extremely stable to simple hydrolysis, the dipeptidyl vc linker, as expected, was found to be quite stable toward hydrolysis in an *in vitro* plasma stability study (133). In this study, a cAC10-vc-MMAE ADC was incubated over 10 days in human or dog plasma at 37°C and less than 2% of the total drug was released as analyzed by liquid chromatography-tandem mass spectrometry (LC-MS/MS). Additionally, no other drug-related molecular species was seen. Over the same time period, free MMAE also did not show any degradation products while incubated in human plasma at 37°C, showing the robust nature of the chemical structure of the auristatin payload.

The chemical stability of the *pabc* spacer moiety was first reported by Katzenellenbogen *et al.* (154) who designed this spacer for use in small molecule prodrug therapies. They reported the *in vitro* stability of a prodrug with this *pabc* spacer to have a  $t_{1/2} \sim 40$  h at 25°C (0.05 M Bistris, pH 6.9). The *pabc* spacer was found to be stable under physiological conditions similar to vc-*pabc*-Dox conjugates from Bristol-Myers Squibb (155,156). Presumably, differences in the solvent accessibility and/or contact surfaces of the payload with the larger antibody conjugates compared to small molecule prodrugs may be attributed for the longer stability in payload release for ADCs using *pabc*.

The reactive moiety on the linker also has a role to play in the chemical stability of the payload-linker. Alley *et al.* (60) compared the stabilities of mc-MMAF and bac-MMAF on mAb 1F6-C4v2. The mc-MMAF conjugation to thiols on the reduced mAb resulted in thioether-succinimidyl ring linkages while the bac-MMAF conjugation resulted in thioether-acetamido linkages. Thioether-succinimidyl ring linkages (aka maleimide linkages) are known to undergo retro-Michael addition and succinimidyl ring-opening hydrolysis (157,158). Deconjugated maleimide payload-linkers in the presence of excessive amounts of circulating reactive thiols (*i.e.*, serum albumin, glutathione) are prone to adduct formation. For the 1FV-C4v2-mc-MMAF conjugate in an *in vivo* assay, payload was found to release with a half-life of 7 days. This correlated with an *in vivo* mouse study of cAC10-vc-MMAE with a release half-life of 6 days (153). The acetamido linkage was more stable *in vivo* with no drug release during the 14 day assay. This increased stability is presumably due to lack of a deconjugation pathway by the thioether-acetamido linkage. It should be noted that these are *in vivo* studies and may involve more than just chemical stability. Baldwin *et al.* (159) investigated the retro reaction of maleimide-thiol conjugates using an *in vitro* system consisting of phosphate buffer at pH 7.4 with and without various thiol compounds (*e.g.*, glutathione). They used model systems of 4-mercaptophenylacetic acid, N-acetylcysteine or 3-mercaptoproprionic acid reacted with N-ethylmaleimide and followed the kinetics of deconjugation

with  $^1\text{H}$ -NMR and HPLC. These model systems showed half-lives of retro reaction from 19 to 337 h and were dependent on the concentration of thiol reductant.

**Maytansinoids.** The chemical stability of the majority of the relevant maytansinoid payload analogs revolves around the *N*-acyl-*N*-methyl-L-alanyl ester moieties at the C3 ring position (see structures of DM1 and DM4). It was found during the synthesis of various analogs that this C3 ester linkage is susceptible to elimination under mild basic conditions ( $\text{pH} > 9$ ) (95) to form the  $\alpha,\beta$ -unsaturated maytansine derivative. Interestingly, the C3 ester was found to be stable to a wide panel of commercially available esterases and lipases in terms of enzymatic hydrolysis. Under oxidative stress conditions (substoichiometric concentrations of hydrogen peroxide at  $37^\circ\text{C}$ ,  $\text{pH}$  5.5 or  $\text{pH}$  7.4 buffer) (160), the ansa macrolide ring exhibited no signs of oxidation. When the C3 ester of the payload contains a free mercapto or disulfide moiety, it is susceptible to typical thiol chemistries: electrophilic addition, metal-catalyzed oxidation (mercapto) or thiol exchange, reduction (disulfides). Although there are a number of chiral centers on this ansa macrolide payload, no evidence of stereoisomerization has been reported (to our knowledge) while utilized on a maytansinoid ADC.

The chemical stability of the linkers used for maytansinoid-loaded ADCs depends on the type of linker chemistry: a disulfide linkage (*i.e.*, SPDP) or a maleimide linkage (*i.e.*, SMCC) (see Fig. 8). The currently marketed T-DM1 utilizes the non-cleavable thioether (SMCC) linker. This maleimide linker was thought to be susceptible to the retro-Michael addition or the succinimide-ring opening hydrolysis seen in other bioconjugates (157,158). A third degradation pathway was postulated by Fishkin *et al.* (160) where the thioether-succinimide linkage undergoes chemical oxidation followed by sulfoxide elimination under mild aqueous conditions. In this study, Fishkin *et al.* used a model compound, DM1 conjugated to *N*-ethylmaleimide

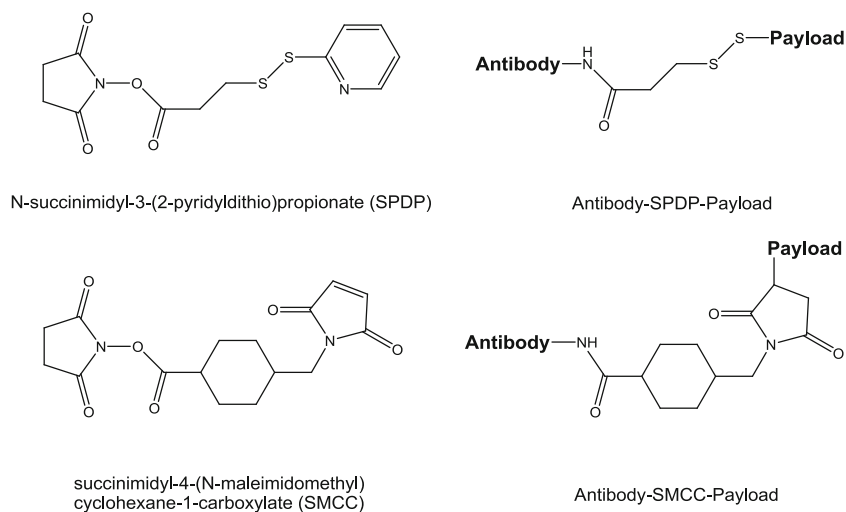
(DM1-NEM). They found that the DM1-NEM was fairly stable in PBS,  $\text{pH}$  7.4 at  $37^\circ\text{C}$  ( $\leq 5\%$  sulfoxide formation after 40 h) but more susceptible in *ex vivo* human plasma ( $37^\circ\text{C}$ ,  $\sim 20\%$  cleavage and sulphonate formation at 40 h). More surprisingly, the unoxidized DM1-NEM showed no apparent retro-Michael reaction in the presence of 1000 eq. of DTT in PBS for 12 h at  $37^\circ\text{C}$ . Under the same conditions, the sulfoxide species yielded 82% free DM1 cleavage product. They found that steric hindrance adjacent to the thioether can attenuate the oxidative degradation pathways for free payload release. One cautionary note is that low levels of autooxidation may have occurred during acetone extraction and sample concentration during HPLC analysis of unoxidized DM1-NEM.

Indeed, care must be taken during sample preparation for payload-linker analysis. This is not only during drug substance and drug product analytical method development, but also during pharmacokinetic analysis. Dere *et al.* (161) described the development of pharmacokinetic assay development for T-DM1. They used orthogonal methods such as ELISA for identifying payload conjugated to antibody and LC-MS/MS for free DM1 payload catabolites. In the sample preparation for the LC-MS/MS assay, extra care was taken to identify various chemically transformed and biotransformed DM1-related species. They noted that  $\sim 0.4\%$  of DM1 in PK samples is releasable in the form of disulfide linkage or by thioether oxidation and sample prep included a reduction and capping step to capture all forms of releasable DM1.

## FORMULATION CONSIDERATIONS

Issues that are faced when developing a formulation for a monoclonal antibody are also relevant for the ADC and possibly exacerbated based on the impact to the structure as discussed earlier. Proteins can be susceptible to degradation when exposed to factors such as temperature,  $\text{pH}$ , agitation

**Fig. 8** Chemical structures of *N*-succinimidyl-3-(2-pyridyldithio)propionate (SPDP) and succinimidyl-4-(*N*-maleimidomethyl)cyclohexane-1-carboxylate (SMCC) linkers before and after typical conjugation reactions with antibody and drug payload.



and light (149,162). When developing a formulation parameters such as protein concentration, pH, buffering species, excipients, solid *versus* liquid dosage form need to be considered. However, an ADC presents some additional challenges to formulation development. Consideration has to be given not only to the overall stability of the mAb/ADC, but also the chemical stability of the payloads and linkers. The conditions that render the optimal stability for the hydrophilic mAb may not be well suited for the hydrophobic drug-linker. The chemical stability of the payloads and linkers are discussed in detail in the previous section. This section will focus on the concerns for developing a formulation for the ADC drug product. For reference, the current commercial products are all lyophilized dosage forms that contain a buffer, stabilizer (*e.g.*, trehalose or sucrose), and surfactant.

An important aspect for ADC's is the increased propensity for self-association driven by the hydrophobic payload(s). Self-association can lead to the formation of aggregates which can impact the potency of the drug, and increase the risk for immunogenicity upon administration (162). High solubility in general is a desirable property of antibody formulations. However, conjugation alters the hydrophilic-hydrophobic balance of the molecule. While antibodies are most soluble in aqueous buffered solutions the hydrophobic linkers and payloads prefer polar organic solvents (163). After conjugation, this duality is incorporated into the construct. The degree of conjugation can cause issues with solubility as the more highly loaded species have the greater tendency to precipitate or aggregate (164,165). Using a hydrophilic linker to balance the hydrophobic payload as part of the design of the ADC has helped mitigate this effect. Highly water-soluble hydrophilic linkers with either a negatively charged  $\alpha$ -sulfonic acid group or a polar polyethylene glycol chain enabled an increase in solubility of antibody-maytansinoid conjugates (166,167). These hydrophilic linkers allowed for a higher DAR than the hydrophobic N-Succinimidyl 3-[2-pyridyldithio]-propionate (SPDP) and succinimidyl 4-(N-maleimidomethyl)cyclohexane-1-carboxylate (SMCC) linkers (Fig. 8), and were shown to be either equal or more cytotoxic. BR96 immunoconjugate aggregation was inhibited by the use of highly branched, hydrophilic methoxytriethylene glycol linkers. This enabled the synthesis of conjugates with molar ratios up to 16 mol of doxorubicin (168). Solubility of a highly hydrophobic ADC can generally not be significantly impacted by formulation approaches.

If the conjugation of the payload is through a surface-exposed lysine, then the reduction in net surface charge can also cause solubility issues due to modification of the electrostatic properties (147). It has been well documented in mAbs that electrostatic surface potential distribution can have an effect on the solubility, thermal stability and viscosity behavior (169–171). Solubility in these situations can in some cases be modulated by changes in ionic strength, pH, and by the addition of certain excipients (*e.g.*, arginine).

The key to identifying a stabilizing formulation for an ADC would be to identify the optimal pH and buffer species that balances the various routes of degradation for the mAb, and for the linker-payload if these too are susceptible. Subsequently, formulating to inhibit aggregation can generally follow the same principles as for a mAb alone, for reviews see (172–174). Typical general formulation excipients/additives that can be included for stabilization of structure and the prevention of aggregation include sugars and polyols, amino acids, amines, polymers, and surfactants (172). More specifically, the examination of higher order structure using various techniques as discussed earlier shows that ADCs are both conformationally (reduced structural stability) as well as colloiddally (increased hydrophobicity) destabilized compared to the mAb. Since the degree of destabilization is related to DAR, the higher DAR species in a pool will drive the overall stability of the product (44). Formulation efforts would therefore require structure stabilizers (*e.g.*, sucrose) in low ionic strength buffers, along with a surfactant to reduce impact of hydrophobicity and interfacial stress (44). [This is confirmed by Adem *et al.*, (132) who show that cys-conjugated MMAE ADC when formulated in high ionic strength buffer, resulted in significantly faster rates of aggregation (and fragmentation) compared to low ionic strength buffer. The predominant aggregating and fragmenting species were of high DAR (6,8)].

The question of optimal dosage form arises next. The ADCs that have gained approval (gemtuzumab ozogamicin, brentuximab vedotin, T-DM1) are all lyophilized products. While this is a small data set, it is quite likely that the lyophilized dosage form will be the preferred model for ADCs. A properly designed lyophilized formulation can enhance the stability of the product significantly while eliminating risk of degradation (physical or chemical). Although liquid dosage forms for mAbs are common, the risk for free drug (payload) formation would seriously preclude the development of a liquid product for the ADC. From a handling safety perspective, opinions differ whether accidental exposure risk to a healthcare provider is greater from a broken vial of liquid or lyophilized powder.

Development of lyophilized dosage forms is well understood in the industry and will not be covered here. The general rules for stabilization of proteins in lyophilized form apply [see *e.g.*, (175)]. A few particular aspects of ADCs are worth mentioning. ADC doses are currently low, of the order of a few mg/kg (*e.g.*, brentuximab vedotin 1.8 mg/kg; T-DM1 3.6 mg/kg; gemtuzumab ozogamicin 9 mg/m<sup>2</sup> or approx. 18–27 mg total) (21,22,51), which in turn implies that the amount in a vial will be low and/or relatively dilute solutions will be lyophilized (*e.g.*, estimated as brentuximab vedotin 5 mg/mL; ado-trastuzumab emtansine 20 mg/mL). At these low protein concentrations, proteins do not contribute to cake structural stability. This would imply that crystallizing additives such as mannitol would be required to obtain good cake



structure while running economical lyophilization cycles. Use of sucrose or trehalose added as cryoprotectants to the ADC drug substance (as discussed earlier) is compatible with lyophilization of drug product. Crystallizing additives may then be added at the DP stage, prior to lyophilization. It is likely that with site-specific conjugates, and for low DARs, the doses will be higher, particularly as the therapeutic index widens. At a load of about 30–50 mg/mL the ADC may itself aid in lyophilization by contributing to cake structure and eliminating the need to add a crystallizing excipient.

## PROCESS CONTROL

The process of mAb (drug substance intermediate, DSI) production is well understood and widely reported. On the other hand, the process for production of ADCs (drug substance, DS and drug product, DP) requires some special considerations, however, the focus of this section will be on the production of DP. The linkers and payloads are generally chemically synthesized or produced from semi-synthetic processes (and outside the scope of this review). However, these latter intermediates as well as the ADCs, have to be produced and handled in facilities specifically designed for high potency drugs. The facility design has to be supplemented with procedures for containment, protection of personnel, protection of product, environmental monitoring as well as (equipment) cleaning. Some of these processing challenges have been discussed by Ducry (176). We discuss below, some related process and use aspects which can have a significant impact but which have not been specifically examined in relation to ADCs.

### Cleaning Requirements

As a consequence of the cytotoxic nature of the ADC product, special containment and handling procedures and facilities are required for the protection of the product as well as the operators. Payloads are produced synthetically or semi-synthetically while the linkers are chemically synthesized. The conjugation reaction is carried out in glass or stainless steel reactor to enable use of organic solvents. The facility for production of ADCs is designed to handle highly potent compounds using closed vessels, isolators, rooms with appropriate air locks and pressure differentials, personnel protective equipment, and cleaning, decontamination and spill clean-up procedures. The ADC drug product has to be produced in similar high containment facilities including lyophilizers with appropriate ventilation as well as cleaning controls. High containment facilities often use peroxide based techniques (*e.g.*, vaporized hydrogen peroxide or hydrogen peroxide gas plasma) to decontaminate/sterilize the isolators or glove boxes where the cytotoxic material is handled for weighing,

pumping, filtration *etc.* Such systems are qualified and have limits on residual peroxide. It is known that peroxides (and other sanitizing agents) can be transferred through the vapor phase into open products (177) resulting in oxidative degradation of proteins. Wang *et al.* (177) also showed that the nature of the impact of sanitizing agent varied with NaOCl causing instantaneous oxidation while H<sub>2</sub>O<sub>2</sub> caused slow oxidation. Therefore it is essential that a risk assessment be performed and the tolerance limits for peroxide contamination in the ADC product be established, to ensure that worse-case exposure in the isolators does not lead to degradation of the product. Of course, proper process design with appropriate aeration cycles and venting times is essential to reduce the contamination levels to acceptable limits in all these technologies. Procedures for handling of components (*e.g.*, bags of stoppers) or cleaning of conveyer belts should preclude use of oxidative sanitizing agents for the same reason. In a contamination event, it is unlikely that the whole product batch would be contaminated uniformly, resulting in erratic release and stability behavior leading to difficulty in establishing cause in later investigations. While the above precautions around sanitizing agents are generally applicable, use of high containment facilities for ADCs makes it particularly important that the development scientists are aware of this risk.

### Storage and Transport of Intermediates

Due to the type of facilities required to produce ADC drug substance and drug product (as discussed above), it is generally quite likely that transport of the mAb intermediate to the conjugation facility and then subsequent transport of the ADC drug substance to the drug product facility would be required. Long-term storage of the mAb and the ADC drug substance in container/closure systems that enable these logistics must be considered. The established scale of bioreactor facilities in many companies is such that the mAb itself is produced in significantly larger amounts than required per batch of the ADC DS. The mAb has to therefore be stored frozen in long-term storage and metered out for ADC production. Frozen storage of mAbs has been extensively discussed in the literature and the formulation knowledge as well as technology is well established (178,179). It is preferable that the mAb is formulated in a buffer and cryoprotectant system that is compatible with the subsequent conjugation step. A surfactant is preferably avoided at this stage. Histidine buffer is commonly used for stabilizing mAbs. It is a weak nucleophile and can interfere in amino conjugations. Phosphate buffers have also been used to formulate mAbs and are generally compatible with most conjugation reaction systems. Common cryoprotectants include sucrose or trehalose, which also tend to be quite compatible with most conjugation chemistries and therefore do not require removal prior to conjugation. Sucrose based systems can be stored for the



long-term at  $-20^{\circ}\text{C} \pm 5^{\circ}\text{C}$ . However, trehalose or high concentration mannitol systems are preferably stored at  $-40^{\circ}\text{C}$  or lower, due to the risk for crystallization of the cryoprotectant and subsequent aggregation of the mAb (180) (181). This latter aspect holds true for the ADC DS also. Once the conjugation is completed, the final ultrafiltration/diafiltration UF/DF step is used to convert the ADC into the formulation intended for the DP. This is likely to be the preferred buffer with a cryo/lyoprotectant and a nonionic surfactant.

The choice of the container closure system for the mAb is determined by the combination of temperature of storage as well as the logistics of the conjugation step. Similarly, storage of ADC drug substance has to take into consideration the batch-size as well as operational flow at the DP facility. Celsius® bags made of ethylvinyl acetate (EVA) are available in two sizes that can accommodate between 2 L and 16 L and can be frozen and thawed in a reproducible manner using an active freeze/thaw unit (182). Smaller-sized bags of the same material are available but must undergo passive processing. EVA bags are preferably stored above  $-55^{\circ}\text{C}$  due to the risk for brittle fracture below this temperature. Bags made of low density polyethylene (LDPE) are also available and can be used with passive (simple freezer) systems. More flexible options are offered by storage in polymer bottles such as polytetrafluoroethylene (PTFE), high density polyethylene (HDPE), polycarbonate (PC) or polyethylene terephthalate glycol-modified (PETG) with various sizes being available in pre-sterilized configuration (except PTFE). PTFE, HDPE and PC bottles have a very wide operational temperature range (down to  $-90^{\circ}\text{C}$ ) while PETG has a brittleness temperature of  $-40^{\circ}\text{C}$ . Use of bottle makes it especially important that proper procedures be defined for each step such as filling, freezing, re-torquing of cap after freezing, handling and storage, to ensure reproducibility and sterility and maintain integrity. Thawing is preferably performed at room temperature (to enable rapid thaw) so as to prevent cryo-concentration and ionic strength (as well as potentially pH) driven variations to cause degradation (178,183,184).

### Stability of ADCs in IV Infusion Bags

Most ADCs are intended for oncology indications and are expected to be administered *via* intravenous infusion in a health-care provider setting. Dilution of the ADC drug product into an infusion solution (*e.g.*, normal saline 0.9% NaCl, 5% dextrose, Lactated Ringers, *etc.*) is likely to enable the infusion at a defined rate. An intravenous push or bolus injection is not allowed with any current ADC product for safety reasons since an infusion can be terminated if an adverse event is observed. The brentuximab vedotin prescribing information (21) recommends dilution of the reconstituted drug product so as to achieve a concentration between 0.4–1.8 mg/mL in the infusion solution. The infusion solution should be dosed

immediately or stored at  $2-8^{\circ}\text{C}$  and used within 24 h of reconstitution. Freezing of the infusion solution is contraindicated. T-DM1 infusion solution is prepared by dilution of reconstituted solution only into normal saline diluent (22). Use of 5% dextrose as a diluent is not allowed, in accordance with the instructions for parent mAb Herceptin. The infusion solution is recommended to be used immediately or stored at  $2-8^{\circ}\text{C}$  for a maximum of 4 h prior to use. Again, freezing is contraindicated. The difference in use period of the prepared infusion solution may relate to the amount of data provided for microbiological quality verification (185). For both of the above products, addition of other drugs into the infusion solution is not allowed. Co-administration *via* the same bag or infusion system is generally disallowed due to lack of knowledge of chemical or physical compatibility with other drugs. On the other hand, it is noted through *in vitro* studies that DM1 in Kadcyla is mainly metabolized by CYP3A4 enzyme and to a lesser extent by CYP3A5 enzyme. Therefore, it is recommended that concomitant therapy (also unfortunately called co-administration but more appropriately called concomitant therapy or use) with strong inhibitors of CYP3A4 should be avoided as it would potentially increase the exposure to DM1 and toxicity, although the PK Is not expected to be impacted. A similar risk is associated for Adcetris ((29,30).

The content of the Dosing and Administration Section of the Prescribing Information is defined by the FDA (186). Short-term stability studies are required to document the suitability of the dosing and administration instructions provided in the examples above. These studies are designed based on the clinical dosing protocol and cover the range of dilution concentrations expected in the clinic. Development studies help to identify infusion solutions that are suitable or contraindicated (186–188). Prepared infusion solutions covering the targeted concentration range are then held for 24 or 48 h at  $2-8^{\circ}\text{C}$  and/or room temperature and analyzed to document the use period of the infusion solution. Impact of light has to be incorporated (189). While such studies are fairly routine, some particular aspects are worth mentioning in the context of ADCs.

Dose levels for ADCs can be quite low, especially during Phase I clinical studies. This may require significant levels of dilution of the ADC drug product leading to risk of adsorptive losses combined with difficulties in analysing and assessing the quality parameters at the low concentrations. Hydrophobicity of the payload can enhance adsorption onto plastics which may be ameliorated to some extent by the addition of surfactants. However, dilution for dosing reduces the level of stabilizing excipients and excipients that prevent surface adsorption, particularly the surfactant, and reduces the stability of the ADC increasing the risk for aggregation and particle formation (188,190). Dosage and administration confirmation studies can be designed to determine the highest dilution levels that may be reproducibly used for dose administration. High

salt concentrations can enhance hydrophobic interactions and impact solubility. Although normal saline is a relatively low ionic strength solution, non-specific electrostatic screening can potentially lead to reduced solubility and aggregate and particulate formation, especially of high DAR species. Apart from the above physical liabilities, chemical liabilities may or may not be enhanced in the infusion solution, depending on the properties of the components of the ADC. No specific reports are available but the critical attributes of the ADC (binding ability and affinity of mAb, chemical stability of payload, amount of free payload, DAR) have to be examined. It is known that certain payloads (*e.g.*, calicheamicin) are sensitive to light. In these cases, the dose preparation and administration instructions have to be qualified with appropriate precautionary instructions (*e.g.*, preparation under shielded fluorescent lighting and shrouding of infusion bag with a UV light protective cover [see *e.g.*, (51)]).

## QUALITY AND REGULATORY ASPECTS

The complexity of the ADC construct requires that all components are appropriately characterized and controlled prior to assembly. The payload and linker are individually controlled as (small molecule) APIs with the same requirements for characterization of structure, purity, impurity profile, degradation *etc.* (191). Impurities above 0.1% threshold would have to be identified and above 1% would have to be qualified per International Conference on Harmonisation guidance ICH Q3A(R2) (192). How would the ICH Q3B(R2) (193) be applied to the drug product (*i.e.*, to the ADC DS or the final product)? For oncology products, ICH S9 may be used to assess and justify the risk from impurities (194). The blanket application of the ICH Q3A qualification and identification limits for impurities (from linker and payload) should however be re-considered in light of the fact that only a fraction of the linker and payload APIs are present in the final product. The fate of these impurities, whether they are carried over or cleared during conjugation, must also be considered. Conjugatable impurities can potentially be carried over into the final drug substance and therefore the safety risk they pose must be understood and limits set. However, the likelihood that a conjugatable impurity is “more” toxic than its cytotoxic parent is small. Impurities with genotoxic potential may require stricter controls, but considering the generally life-threatening indications and nonchronic usage of ADCs, limits higher than the threshold for toxicological concern (of 1.5 µg/day exposure) may be used as allowed for in the European Medicines Agency EMA guidance (195). Further discussion on the risk assessment strategy is available in a forthcoming publication by Jones *et al.* (196).

Assays, specifications, preferred storage conditions and shelf-life has to be set for each component - the linker, the

payload and of the intermediate if these linker and payload are reacted and stored prior to the conjugation step. Changes to the linker and payload processes can potentially be managed, but the impact (or lack thereof) on the quality of the ADC has to be shown (197).

The current practice for ADCs are to understand and control the parent mAb per current standards for mAb products including identity, purity, degradation products, size and charge heterogeneity, and process-related impurities. A binding or potency assay to reflect biological activity is a requirement, typically a binding ELISA. Furthermore, if the mAb has ADCC or CDC activity, additional assays may be required for the mAb, the ADC or both.

The ADC requires a host of additional assays, including the DAR, drug load distribution, free payload, free antibody, size heterogeneity and process-related impurities. The presence of free antibody is theoretically a concern due to potential for competition for receptor occupancy with the ADC but the likelihood of saturating the antigen sites before encountering toxicity is very small. A potency assay for the cytotoxicity of the payload would be expected. The assays required for release and stability depend on the ADC product and process, and typically include free payload, aggregates, size heterogeneity and other degradation products (where present). Some of these parameters may have to be monitored on stability also. The presence/generation of free payload could have an impact on safety. Setting specifications for this and other quality attributes may be a challenge since manufacturing experience at licensure is generally limited and a direct link between the possible range of quality parameters and clinical outcomes is difficult to accomplish. A possibility may be to use preclinical or *in vitro* models to test impact of conjugation-related quality parameters and help guide specifications.

The ADC must also be well-characterized by a variety of methods. Characterization should explore the structural and functional elements of the ADC. Demonstration that conjugation does not impact antigen binding is a critical component of the characterization. Comparability studies for ADCs should follow a strategy similar to that for any biologic, with side-by-side characterization of pre- and post-change lots. Process-related impurities must be characterized and controlled, while any co-solvents need to be tested and demonstrated to be cleared per ICH Q3C.

Wakankar *et al.* (129) have provided a review of the analytical methods for physicochemical characterization of ADCs, while challenges around the bioanalytical aspects, including immunogenicity assays, have been reviewed by Xiao *et al.* (198). Analytical characterization and regulatory aspects have also been reviewed by Harris *et al.* (197).

An important regulatory aspect of the ADC is the designation of regulatory starting materials. Most payloads are derived from synthetic or semi-synthetic (chemically modified fermentation product) processes. Linkers may be chemically

synthesized. In this case, the identification of the point at which the materials are designated as starting materials and need to be under proper current Good Manufacturing Practices (cGMP) and change management controls as well as reporting requirements, becomes important (ICHQ7; also see ICH Q11) (199,200). Actual designation is best done in discussion with the regulatory agency (201). It is also recommended to us as many combinations of linker/payload and mAb as possible during clinical development to create a database of quality parameters that get tested in clinic.

The review of an ADC CMC section submission at the US FDA is a collaborative effort that includes reviewers from the Office of Biotechnology Products (OBP) and Office of New Drug Quality Assessment (ONDQA) along with the Biotech Manufacturing Assessment Branch (191). The OBP focuses on the characterization, comparability, impurities, testing and specifications for the mAb, DS and DP. The ONDQA focuses on the starting materials and intermediates and on characterization, testing and specifications of linker, payload, DS and DP. The IND (Investigational New Drug) Module 3S is preferably organized as 3 separate DS sections/folders, one each for the mAb (DSI), linker/payload intermediates, and for the final DS (191). An interesting divergence in the traditional approaches to the validation section (3.S.3.5) arises in that biologics (*i.e.*, mAb and therefore the ADC) usually provide process validation data in a BLA (Biologics License Application) while small molecules (*i.e.*, linker and payload) provide a validation protocol in the NDA. A risk-based strategy to the process validation should be considered in terms of whether to use validated components in the validation of the ADC.

## DISCUSSION AND CONCLUSIONS

This review highlights some of the product developmental aspects of antibody-drug conjugates. With the recent approval of two ADCs, these therapeutic modalities are becoming an increasing part of many major biopharmaceutical drug pipelines. Three distinct elements (payload, linker and mAb) have to combine synergistically to lead to a successful ADC. A number of new technologies are being explored in this field within these three elements. As the understanding grows, the development is beginning to move away from empiricism to rational design, to improve safety, efficacy and thus therapeutic index, as well as PK-PD. Quality and regulatory aspects are also evolving but the key to development is a deep understanding of the physicochemical characteristics studied by a variety of techniques. Advances in ADC design, improved stability *in vitro* and *in vivo*, and appropriate targeting of diseased tissue is allowing the clinical application of ADCs to accelerate.

However, there are many facets of the technology that require further development and understanding. The ability to understand the impact of isotype and the site and type of conjugation on PK-PD or efficacy without resorting to animal studies would remove some of the empiricism. Furthermore, the relevance of animal models to human response is not always clear. The same goes for the development of novel linkers and payloads and their impact on structure and functional properties of the ADCs, including specificity, toxicity, solubility and stability. Improved process understanding may allow broader strategies for product control beyond operating the process in a narrow range. A tighter control on heterogeneity through improvements in site-specific conjugation technologies will help improve safety and efficacy. Current products and a number of those in development are lyophilized presentations. Can excipients be identified or developed to allow a liquid product? Continuing advances in characterization methods will be required as novel combinations are created, tested and produced. Finally, the question remains if the technology can find utility beyond oncology.

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